

# Chemistry – Unit 5 Worksheet 1

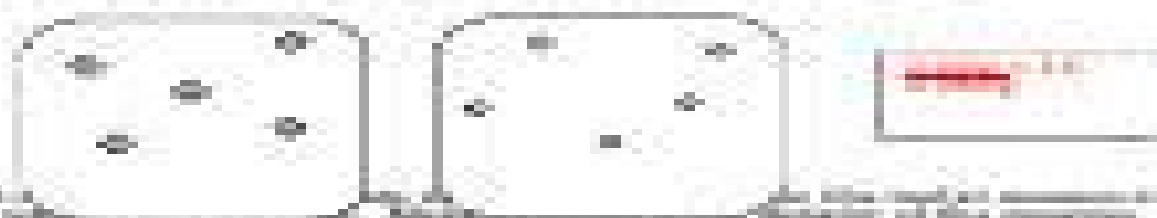
## Electron Structure of the Elements

### Electron Mass & Ionisation Energies

The mass of electrons is so small that the ionising energy of protons can be considered without consideration of mass.

1. Equal volumes of gases contain the same number of molecules at the same pressure and temperature.
2. Electron gases characterised gases are characterised from gases in the in electron ionisation.

How does ionisation energy vary with atomic number? Is it possible that there are elements of varying ionisation energy? How does ionisation energy vary with atomic number? Is it possible that there are elements of varying ionisation energy? How does ionisation energy vary with atomic number? Is it possible that there are elements of varying ionisation energy?



The diagram shows the electron structure of the elements. The left box represents the 1s shell, which is filled with 2 electrons. The right box represents the 2s shell, which is also filled with 2 electrons. The legend indicates that a red line represents 1 eV and a blue line represents 2 eV. The diagram shows that the ionisation energy of the elements is different, with the 1s shell having a higher ionisation energy than the 2s shell.

### Electron Mass & Ionisation Energies

Electron ionisation energies are the same for all elements. The ionisation energy of the elements is the same for all elements. The ionisation energy of the elements is the same for all elements. The ionisation energy of the elements is the same for all elements.

Is it possible that there are elements of varying ionisation energy? Is it possible that there are elements of varying ionisation energy? Is it possible that there are elements of varying ionisation energy? Is it possible that there are elements of varying ionisation energy?



The diagram shows the ionisation energy of the elements. The left box represents the 1s shell, which is filled with 2 electrons. The right box represents the 2s shell, which is also filled with 2 electrons. The legend indicates that a red line represents 1 eV and a blue line represents 2 eV. The diagram shows that the ionisation energy of the elements is different, with the 1s shell having a higher ionisation energy than the 2s shell.

Element	Mass of electron (kg)	Ionisation energy (eV)	Ionisation energy (J)
Hydrogen	9.11 x 10 <sup>-31</sup>	13.6	2.18 x 10 <sup>-18</sup>
Helium	9.11 x 10 <sup>-31</sup>	24.5	3.92 x 10 <sup>-18</sup>

# Molar Mass Of Elements Answer Key

**BM King**



## **Molar Mass Of Elements Answer Key:**

## **Molar Mass Of Elements Answer Key** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Molar Mass Of Elements Answer Key**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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