

**Molarity Practice Worksheet**

Find the molarity of the following solutions: Molarity =  
mole/Liters

**1 liter = 1000 mL**

- 1) 2.5 moles of NaCl is dissolved to make 0.050 liters of solution.
  
  
  
  
  
  
  
  
  
  
- 2) 3.0 moles of AgCl is dissolved in enough water to make a 2.0 liter solution
  
  
  
  
  
  
  
  
  
  
- 3) 4.2 moles of lithium fluoride are dissolved in 2 liters of solution.
  
  
  
  
  
  
  
  
  
  
- 4) 18 grams of hydrochloric acid, HCl is dissolved to make 1 liter of solution.

# Molality Worksheet With Answers

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A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Lincoln Learning Solutions text.

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