

CHAPTER 8 REVIEW

Chemical Equations and Reactions**Teacher Notes and Answers****Chapter 8****SECTION 1****SHORT ANSWERS**

1. a. 2

b. 2

c. 2

d. 1

e. 2

f. 1

2. a. 2, 2, 2

b. a. 12 atoms

b. 12 atoms

c. 12 atoms

d. 1 + 1 = 2 H^+ atoms3. $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{NaHSO}_4 + \text{H}_2\text{O}$ or $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow 2\text{NaHSO}_4 + \text{H}_2\text{O}$ 4. $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$ or $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$ 5. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the products6. a. When solid calcium hydroxide reacts with hydrochloric acid, it forms a clear solution. When solid calcium hydroxide reacts with nitric acid, a clear solution forms.
b. When solid calcium hydroxide reacts with hydrochloric acid, it forms a clear solution. When solid calcium hydroxide reacts with nitric acid, it forms a clear solution.7. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the products
c. HNO_3 and H_2 are the products
d. HNO_3 and H_2 are the products
e. HNO_3 and H_2 are the products
f. HNO_3 and H_2 are the products8. a. HNO_3 b. HNO_3 **SECTION 2****SHORT ANSWERS**

1. a. 2

b. 2

c. 2

d. 1

2. a. 2

b. 2

c. 2

3. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$ b. HNO_3 and H_2 are the productsc. HNO_3 and H_2 are the productsd. HNO_3 and H_2 are the products4. a. HNO_3 and H_2 5. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the productsc. HNO_3 and H_2 are the productsd. HNO_3 and H_2 are the productse. HNO_3 and H_2 are the productsf. HNO_3 and H_2 are the products6. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the productsc. HNO_3 and H_2 are the productsd. HNO_3 and H_2 e. HNO_3 and H_2 7. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the productsc. HNO_3 and H_2 are the products8. a. HNO_3 and H_2 are the products**SECTION 3****SHORT ANSWERS**1. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the products2. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the products3. a. HNO_3 b. HNO_3 c. HNO_3 4. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the productsc. HNO_3 and H_2 d. HNO_3 and H_2 e. HNO_3 and H_2 f. HNO_3 and H_2 5. a. $2\text{HNO}_3 + \text{Ca} \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$
b. HNO_3 and H_2 are the productsc. HNO_3 and H_2 d. HNO_3 and H_2 e. HNO_3 and H_2 f. HNO_3 and H_2

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