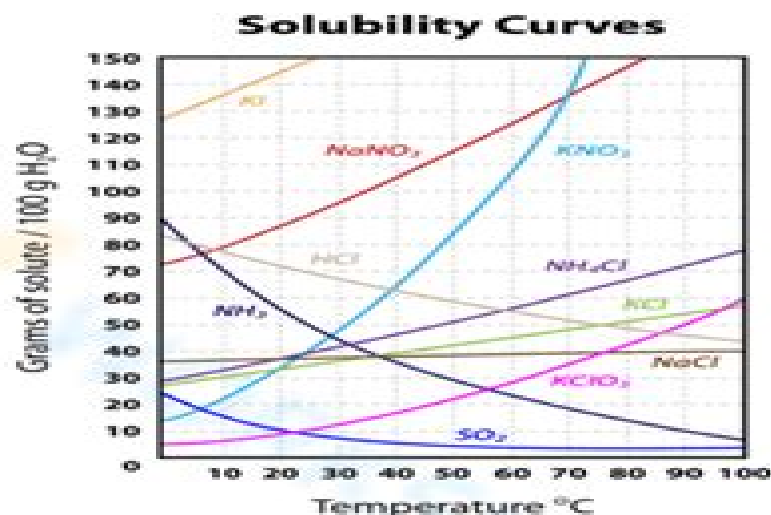


WS 6.1: Interpreting Solubility Curves

Directions: Use the Solubility Curve Table below to answer all questions. So you do not have to keep flipping your paper over you might want to put the solubility curve on your phone or computer screen when answering the questions on the back. All questions assume the amount of water is 100 mL.



- What is the solvent in this Solubility Curve?
water
- Saturation points are given per how much solvent?
100 ml or 100 g
- What is the saturation point of NaNO₃ at 10°C, 40°C, & 80°C?
@ 10°C = ~79 g
@ 40°C = ~105 g
@ 80°C = ~146 g
- How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
@ 10°C = ~70 g
@ 30°C = ~45 g
@ 90°C = ~10 g
- What is the maximum amount of KCl that can be dissolved in water at 20°C, 50°C, & 90°C?
@ 20°C = ~33 g
@ 50°C = ~42 g
@ 90°C = ~53 g
- If you want to make a saturated solution of NaCl how many grams would you dissolve in 100 grams of water @ 80°C?
About 40 grams
- Except for KI What substance is the most soluble in water at 20°C?
NaNO₃
- Except for KI, what substance is the most soluble in water at 72°C?
KNO₃
- Look up The boiling points for SO₂, HCl and NH₃ and write them below:
B.P. SO₂ = -10°C
B.P. HCl = -85°C
B.P. NH₃ = -33°C
- Based on the boiling points you looked up, are SO₂, HCl & NH₃ solids, liquids, or gases at 0°C? (and anywhere else on the solubility curve)
gases

Interpreting Data From Solubility Curves

Lei Shi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Shi'.

Interpreting Data From Solubility Curves:

This book delves into Interpreting Data From Solubility Curves. Interpreting Data From Solubility Curves is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Interpreting Data From Solubility Curves, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Interpreting Data From Solubility Curves
- Chapter 2: Essential Elements of Interpreting Data From Solubility Curves
- Chapter 3: Interpreting Data From Solubility Curves in Everyday Life
- Chapter 4: Interpreting Data From Solubility Curves in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Interpreting Data From Solubility Curves. This chapter will explore what Interpreting Data From Solubility Curves is, why Interpreting Data From Solubility Curves is vital, and how to effectively learn about Interpreting Data From Solubility Curves.

3. In chapter 2, the author will delve into the foundational concepts of Interpreting Data From Solubility Curves. The second chapter will elucidate the essential principles that must be understood to grasp Interpreting Data From Solubility Curves in its entirety.

4. In chapter 3, the author will examine the practical applications of Interpreting Data From Solubility Curves in daily life. The third chapter will showcase real-world examples of how Interpreting Data From Solubility Curves can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Interpreting Data From Solubility Curves in specific contexts. This chapter will explore how Interpreting Data From Solubility Curves is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Interpreting Data From Solubility Curves. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Interpreting Data From Solubility Curves.

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