Balance the following equation:

ne following equation: 
$$\frac{8}{13}$$

$$2\left(\frac{2}{2}C_4H_{10} + \frac{65}{5}O_2 \rightarrow \frac{4}{5}CO_2 + \frac{5}{5}H_2O\right)$$

1. How many moles are in 4.52 x 10 30 molecules of O, ?

4.52×1030 molecules bozgodolecules

2. If you have 3.5 moles of O2, how many moles of H2O will you make?

If you have 50.0 g of C<sub>4</sub>H<sub>10</sub>, how many moles of O<sub>2</sub> will you need?

4. If you have 100.0 g of O2, how many grams of CO2 will you make?

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#### Honors Chemistry Semester 2 Final Review Answers

web apr 16 2015 have you ever thought of a chicken egg as one big cell of course it is made up of many many actual cells but you can use it as a model to explore how different fluids get transferred from osmosis eggs center for nanoscale science pennsylvania - Apr 03 2023

web measure each egg around at the widest point using a tape measure or a piece of string to use a piece of string take a piece of string and wrap it around the egg once mark with your finger the length of string that is needed to go around the egg once and measure it using a