Name: \_\_ Date: Mods:

#### Honors Chemistry Midterm: Review Packet

#### PART 1

Write in the number of significant figures for each of the following measurements:

3 0.00290 g 4 50.00 m 2 0.070 kg 1

- 2. Using the following masses: 0.0562 kg, 124.213 g and 1635 mg. 

  The answer should be reported as: 

  9

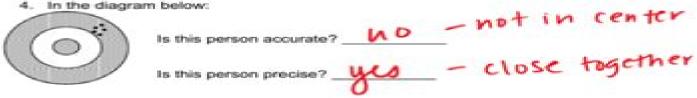
  1.635g

  1.435g
- a) A 5.75 mL sample of mercury has a measured mass of 77.05 g. The density is 13.4 3/mL.

b) Mercury's accepted density is 13.53 g/mL. The % error in your measurement is: . 961 70

acture ×100 113,633/mz -13,43/mul ×100

In the diagram below:



- Identify the following as a chemical change (CC), physical change (PC), chemical property (CP). or physical property (PP):
  - a) burning wood CC
- b) mass of wood PP
- c) flammability of wood CP d) chopping up wood PC

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