

Name: _____

Honors Chem Test: Ch. 7

Multiple Choice—1 point each

Identify the choice that best completes the statement or answers the question.

- How many atoms of fluorine are present in a molecule of carbon tetrafluoride, CF_4 ?
 - 1
 - 2
 - 4
 - 5
- Name the compound Al_2S_3 .
 - aluminum sulfate
 - aluminum sulfur
 - aluminum(II) sulfate
 - aluminum sulfide
- Name the compound CF_4 .
 - calcium fluoride
 - carbon fluoride
 - carbon tetrafluoride
 - mono-carbon quadrafluoride
- Name the compound N_2O_5 .
 - dinitrogen oxide
 - nitrogen trioxide
 - nitric oxide
 - dinitrogen trioxide
- What is the formula for diarsenic trisulfide?
 - As_2S_3
 - AsS_3
 - As_3S_4
 - As_3S_3
- What is the formula for sulfur dichloride?
 - SCl
 - SCl_2
 - S_2Cl
 - S_2Cl_2
- What is the formula for hydrochloric acid?
 - HF
 - HCl
 - HClO
 - H_2CO_3
- The molar mass of an element is the mass of one
 - atom of the element.
 - crystal of the element.
 - gram of the element.
 - mole of the element.
- The molar mass of MgI_2 is
 - the sum of the masses of 1 mol of Mg and 2 mol of I.
 - the sum of the masses of 1 mol of Mg and 1 mol of I.
 - the sum of the masses of 1 atom of Mg and 2 atoms of I.
 - the sum of the masses of 1 atom of Mg and 1 atom of I.

Honors Chemistry Chapter Practice Test

Scott C. Dulebohn



Honors Chemistry Chapter Practice Test:

Honors Chemistry Chapter Practice Test Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Honors Chemistry Chapter Practice Test**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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