

Graphing Skills continued

2. How many grams of the parent isotope are left in the sample after three half-lives?
3. Why is the line graph a curve instead of a straight line?
4. If a sample contained 94,000 g of the daughter isotopes, where on the line graph would the sample be shown?
5. The table below shows the radioactive decay of a 10 kg sample of ^{14}C .

Years Passed	^{14}C (g)	Years Passed	^{14}C (g)
0	10,000	34,200	156
5,700	5,000	28,500	812
11,400	2,500	39,900	78
17,100	1,250	45,600	39
22,800	625	51,300	20

Create a graph using the data above. Label the *x*-axis *Number of Half-Lives*. Label the *y*-axis *Number of Atoms*. Plot the decay of ^{14}C in terms of half-lives. Each half-life for ^{14}C is about 5,700 years.



6. About how old is a sample of bone that contains 900 ^{14}C atoms? Mark its position on the line graph. How many half-lives have passed?

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