

Assessment 1**Chapter Test B****Chapter: Chemical Equations and Reactions**

Part 1 In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The precipitation of a slightly soluble solid compound from a solution is called a _____ reaction, reactions that are the formation of a _____.
- gas
 - precipitate
 - redox reaction
 - halogen
- _____ 2. To balance a chemical equation, it is permissible to adjust the _____.
- coefficients
 - subscripts
 - formulas of the products
 - number of products
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is _____.
- solid
 - dissolved in water
 - an acid
 - aqueous
- _____ 4. The tendency for a replacement reaction to occur depends on the _____ between any two elements in the activity series of metals.
- temperature difference
 - valence electrons and mass
 - relative tendency of a less element in the activity series decreases
 - relative tendency of a less element in the activity series increases
- _____ 5. The coefficients in a chemical equation _____.
- indicate the number of moles of each substance that react
 - show the number of grams of each substance that react
 - are the molar masses of the substances
 - show the valence electrons for each atom
- _____ 6. If metal X is more reactive than metal Y in the activity series, then _____.
- X will replace ions of Y in solution
 - Y will replace ions of X in solution
 - Y will react with cold water
 - X will react with cold water

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