

Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry
Section 1: What is Chemistry?

Objectives:

- **Describe** ways in which chemistry is a part of your daily life.
- **Describe** the characteristics of three common states of matter.
- **Describe** physical and chemical changes, and give examples of each.
- **Identify** the reactants and products in a chemical reaction.
- **List** four observations that suggest a chemical change has occurred.

A _____ is any substance that has a definite composition. A _____ is the process by which one or more substances change to produce one or more new substances. The _____ of _____ are the physical forms of matter which are solid, liquid, gas, and plasma.

- _____ have a fixed volume and shape that result from the way their particles are arranged.
- _____ have a fixed volume but not a fixed shape.
- _____ have neither fixed volume nor shape.



Physical changes are changes in which the _____ of a substance doesn't change.

- Changes of state are physical changes.

Chemical changes occur when the _____ of substances change and new substances form.

Evidence of a chemical change:

			
a When acetic acid, vinegar, and sodium hydrogen carbonate, or baking soda, are mixed, the solution bubbles as carbon dioxide forms.	b When solutions of sodium sulfide and cadmium nitrate are mixed, cadmium sulfide, a solid precipitate, forms.	c When aluminum reacts with iron(III) oxide in the clay pot, energy is released as heat and light.	d When phenolphthalein is added to ammonia dissolved in water, a color change from colorless to pink occurs.

Holt Chemistry Nuclear Chapter Test

Roger S. Macomber



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Holt Chemistry Ralph Thomas Myers, 2004 [AP Chemistry For Dummies](#) Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam

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