

## Skills Worksheet

## Concept Review Chapter 5: Section 1

## Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

# Holt Chemistry Concept Review Answers Chap 10

**Minjie Lin**



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*Holt Chemistry* Ralph Thomas Myers, 2004      MCAT General Chemistry Review 2024-2025 Kaplan Test Prep, 2023-07-04  
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*Basic Chemistry* Steven S. Zumdahl, 2004 Description Not Yet Available

**CHEMISTRY** SILBERBERG, 2003

**The Spectator** , 1849

*Life in the Universe, 5th Edition* Jeffrey Bennett, Seth Shostak, Nicholas Schneider, Meredith MacGregor, 2022-05-31 The world s leading textbook on astrobiology ideal for an introductory one semester course and now fully revised and updated Are we alone in the cosmos How are scientists seeking signs of life beyond our home planet Could we colonize other planets moons or even other star systems This introductory textbook written by a team of four renowned science communicators educators and researchers tells the amazing story of how modern science is seeking the answers to these and other fascinating questions They are the questions that are at the heart of the highly interdisciplinary field of astrobiology the study of life in the universe Written in an accessible conversational style for anyone intrigued by the possibilities of life in the solar system and beyond Life in the Universe is an ideal place to start learning about the latest discoveries and unsolved mysteries in the field From the most recent missions to Saturn s moons and our neighboring planet Mars to revolutionary discoveries of thousands of exoplanets from the puzzle of life s beginning on Earth to the latest efforts in the search for intelligent life elsewhere this book captures the imagination and enriches the reader s understanding of how astronomers planetary scientists biologists and other scientists make progress at the cutting edge of this dynamic field Enriched with a wealth of engaging features this textbook brings any citizen of the cosmos up to speed with the scientific quest to discover whether we are alone or part of a universe full of life An acclaimed text designed to inspire students of all backgrounds to explore foundational questions about life in the cosmos Completely revised and updated to include the latest developments in the field including recent exploratory space missions to Mars frontier exoplanet science research on the origin of life on Earth and more Enriched with helpful learning aids including in chapter Think about It questions optional Do the Math and Special Topic boxes Movie Madness boxes end of chapter exercises and problems quick quizzes and much more Supported by instructor s resources including an illustration package and test bank available upon request

Modern NMR Approaches to the Structure Elucidation of Natural Products Antony J. Williams, Gary E. Martin, David Rovnyak, 2016 The first volume in a two part set that discusses contemporary NMR approaches for the structure elucidation of natural products It covers optimized hardware and experimental approaches

Westminster and Foreign Quarterly Review , 1880

**Saturday Review** , 1863

**Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

**Fundamentals of Chemistry: A Modern**

**Introduction** Frank Brescia, 2012-12-02 Fundamentals of Chemistry Third Edition introduces the reader to the fundamentals of chemistry including the properties of gases atomic and molecular weights and the first and second laws of thermodynamics Chemical equations and chemical arithmetic are also discussed along with the structure of atoms chemical periodicity types of chemical bonds and condensed states of matter This book is comprised of 26 chapters and begins with a historical overview of chemistry and some terms which are part of the language of chemists Separation and purification are covered in the first chapter while the following chapters focus on atomic and molecular weights stoichiometry the structure of atoms and types of chemical bonds The molecular orbital MO theory of bonding galvanic cells and chemical thermodynamics are considered next Separate chapters are devoted to MO theory of covalent and metallic bonding orbital hybridization intermolecular forces acids and bases ionic equilibrium calculations and polymers and biochemicals This monograph is intended for chemistry students

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