

Assessment

# Quiz

## Section: Describing Chemical Reactions

In the space provided, write the letter of the term or phrase that best answers the question.

1. The formation of a gas is evidence of
  - a. a chemical change.
  - b. a physical change.
  - c. no change in energy.
  - d. Both (a) and (c).
  
2. Which of the following may be evidence of a chemical change?
  - a. release of energy as heat
  - b. production of sound
  - c. reduction in temperature
  - d. All of the above
  
3. In the word equation, iron + copper(II) sulfate → iron(II) sulfate + copper, a product is
  - a. iron(II) sulfate.
  - b. copper.
  - c. iron.
  - d. Both (a) and (b)
  
4. In the reaction described by the following equation water + energy → hydrogen gas + oxygen gas, energy
  - a. can be considered a reactant.
  - b. is absorbed.
  - c. is released.
  - d. Both (a) and (b)
  
5. In the unbalanced formula equation carbon + hydrogen → methane, the formula CH<sub>4</sub> represents
  - a. one atom of methane.
  - b. five atoms of methane.
  - c. one molecule of methane.
  - d. one gram of methane.

# Holt Chemistry Chemical Equations And Reactions Answers

**Ian Pickup**



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