



LAB: Exothermic reactions

VAS values integrity. Please do your own work.
No cheating.
Use a sharp pencil to do the graph.

1. In this investigation, I am trying to find out will increase in mass of the reactants affect the temperature change of a reaction.

(2)

2. Variables

The independent variable (factor I will change) is mass of calcium oxide.

(1)

The dependent variable (factor I will measure) is the temperature of the water (start and finish temperature).

(1)

Two controlled variables (factors to keep the same) are: volume of water, type of the reactants.

(1)

Why is it important to keep the controlled variables the same?

It is important to keep the controlled variables the same because it is one of the factors to make an experiment a fair test and so it could give reliable, accurate results.

(1)

3. Results

Complete the headings of the results table.

Then complete the results table. Round to the first decimal place, if applicable.

(6)

Mass of calcium oxide added / g	Start temperature of the water / °C	Finish temperature of the water / °C	Temperature rise / °C
1	23.1	25.1	2.0
2	22.9	25.4	2.5
3	23.2	26.7	3.5
4	26.7	28.2	1.5

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L Reisser



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