

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

Holt Chemistry Concept Review Simple Ions

**Kenneth Malcolm Mackay, Rosemary
Ann Mackay**



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Holt Chemistry Ralph Thomas Myers, 2004 Holt Chemistry Holt Rinehart & Winston, 2003-01-24 Reviews Of Modern Quantum Chemistry: A Celebration Of The Contributions Of Robert G Parr (In 2 Vols) Sen Kali Das, 2002-12-09 This important book collects together state of the art reviews of diverse topics covering almost all the major areas of modern quantum chemistry The current focus in the discipline of chemistry synthesis structure reactivity and dynamics is mainly on control A variety of essential computational tools at the disposal of chemists have emerged from recent studies in quantum chemistry The acceptance and application of these tools in the interfacial disciplines of the life and physical sciences continue to grow The new era of modern quantum chemistry throws up promising potentialities for further research Reviews of Modern Quantum Chemistry is a joint endeavor in which renowned scientists from leading universities and research laboratories spanning 22 countries present 59 in depth reviews Along with a personal introduction written by Professor Walter Kohn Nobel laureate Chemistry 1998 the articles celebrate the scientific contributions of Professor Robert G Parr on the occasion of his 80th birthday List of Contributors W Kohn M Levy R Pariser B R Judd E Lo B N Plakhutin A Savin P Politzer P Lane J S Murray A J Thakkar S R Gadre R F Nalewajski K Jug M Randic G Del Re U Kaldor E Eliav A Landau M Ehara M Ishida K Toyota H Nakatsuji G Maroulis A M Mebel S Mahapatra R Carb Dorca Nagy I A Howard N H March S B Liu R G Pearson N Watanabe S Ten no S Iwata Y Udagawa E Valderrama X Fradera I Silanes J M Ugalde R J Boyd E V Lude a V V Karasiev L Massa T Tsuneda K Hirao J M Tao J P Perdew O V Gritsenko M Gr ning E J Baerends F Aparicio J Garza A Cedillo M Galv n R Vargas E Engel A H ck R N Schmid R M Dreizler J Poater M Sol M Duran J Robles X Fradera P K Chattaraj A Poddar B Maiti A Cedillo S Guti rrez Oliva P Jaque A Toro Labb H Chermette P Boulet S Portmann P Fuentealba R Contreras P Geerlings F De Proft R Balawender D P Chong A Vela G Merino F Kootstra P L de Boeij R van Leeuwen J G Snijders N T Maitra K Burke H Appel E K U Gross M K Harbola H F Hameka C A Daul I Ciofini A Bencini S K Ghosh A Tachibana J M Cabrera Trujillo F Tenorio O Mayorga M Cases V Kumar Y Kawazoe A M K ster P Calaminici Z G mez U Reveles J A Alonso L M Molina M J L pez F Dugue A Ma anes C A Fahlstrom J A Nichols D A Dixon P A Derosa A G Zacarias J M Seminario D G Kanhere A Vichare S A Blundell Z Y Lu H Y Liu M Elstner W T Yang J Mu oz X Fradera M Orozco F J Luque P Tarakeshwar H M Lee K S Kim M Valiev E J Bylaska A Gramada J H Weare J Brickmann M Keil T E Exner M Hoffmann J Rychlewski

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1. Understanding the eBook Holt Chemistry Concept Review Simple Ions
 - The Rise of Digital Reading Holt Chemistry Concept Review Simple Ions
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Concept Review Simple Ions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Concept Review Simple Ions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Concept Review Simple Ions
 - Personalized Recommendations
 - Holt Chemistry Concept Review Simple Ions User Reviews and Ratings
 - Holt Chemistry Concept Review Simple Ions and Bestseller Lists
5. Accessing Holt Chemistry Concept Review Simple Ions Free and Paid eBooks
 - Holt Chemistry Concept Review Simple Ions Public Domain eBooks
 - Holt Chemistry Concept Review Simple Ions eBook Subscription Services
 - Holt Chemistry Concept Review Simple Ions Budget-Friendly Options

6. Navigating Holt Chemistry Concept Review Simple Ions eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Concept Review Simple Ions Compatibility with Devices
 - Holt Chemistry Concept Review Simple Ions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Concept Review Simple Ions
 - Highlighting and Note-Taking Holt Chemistry Concept Review Simple Ions
 - Interactive Elements Holt Chemistry Concept Review Simple Ions
8. Staying Engaged with Holt Chemistry Concept Review Simple Ions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Concept Review Simple Ions
9. Balancing eBooks and Physical Books Holt Chemistry Concept Review Simple Ions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Concept Review Simple Ions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Holt Chemistry Concept Review Simple Ions
 - Setting Reading Goals Holt Chemistry Concept Review Simple Ions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Chemistry Concept Review Simple Ions
 - Fact-Checking eBook Content of Holt Chemistry Concept Review Simple Ions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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