

## ➔ CHEMICAL EQUILIBRIUM

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**Q.1:** Justify that chemical equilibrium is dynamic in nature?

**Ans:** In reversible chemical reactions the molecules of reactions collide and convert into products. At the same time the molecules of the products are converted into reactants. When two opposing forces maintain the equal, rates then equilibrium is there and that is dynamic equilibrium in nature.

**Q.2:** When does four types of chemical equilibrium constants for a reaction become equal?

**Ans:** When we have such a reversible reaction, in which the total number of moles of reactants and products are equal then  $K_p = K_c = K_n$ . In other words  $\Delta n = 0$ , where  $\Delta n$  is the difference of number of moles of reactants and products.

**Q.3:** The change of temperature disturbs the equilibrium position and the equilibrium constant of reaction? Justify?

**Ans:** All the reversible reactions are disturbed by changing their equilibrium position and equilibrium constant by disturbing the temperature. Actually change of temperature changes the energy contents of reactants and products.

**Q.4:** Why is solubility of glucose in water is increased by increasing the temperature?

**Ans:** The solubility of glucose in water is an endothermic process. Increase of temperature pushed the system to that side where heat is absorbed, and that is the side where solubility increases.

**Q.5:** What is the nature of solution having  $\text{pH} = 7.5$ ?

**Ans:** The solution with  $\text{pH} = 7$  is neutral and having values more than 7 have greater concentration of  $\text{OH}^-$  and less concentration of  $\text{H}^+$ , so the solution is basic.

**Q.6:** What is buffer capacity?

**Ans:** Buffer capacity is a ability of a buffer to resist the change of  $\text{pH}$ , when a few drops of an acid and base is added from outside. The best buffer is obtained. (i) when  $\text{pH}$  of the buffer is equal to  $\text{pK}_a$  of the acid and the concentration of the salt and acid are equal. (ii) The  $\text{OH}^-$  of the buffer is equal to  $\text{pK}_b$  or the base.

# Holt Chemistry Concept Review Answers Chemical Equilibrium

**L Towne**



## **Holt Chemistry Concept Review Answers Chemical Equilibrium:**

## Unveiling the Magic of Words: A Review of "**Holt Chemistry Concept Review Answers Chemical Equilibrium**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Holt Chemistry Concept Review Answers Chemical Equilibrium**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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