

Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry
Section 1: What is Chemistry?

Objectives:

- **Describe** ways in which chemistry is a part of your daily life.
- **Describe** the characteristics of three common states of matter.
- **Describe** physical and chemical changes, and give examples of each.
- **Identify** the reactants and products in a chemical reaction.
- **List** four observations that suggest a chemical change has occurred.

A _____ is any substance that has a definite composition. A _____ is the process by which one or more substances change to produce one or more new substances. The _____ of _____ are the physical forms of matter which are solid, liquid, gas, and plasma.

- _____ have a fixed volume and shape that result from the way their particles are arranged.
- _____ have a fixed volume but not a fixed shape.
- _____ have neither fixed volume nor shape.



Physical changes are changes in which the _____ of a substance doesn't change.

- Changes of state are physical changes.

Chemical changes occur when the _____ of substances change and new substances form.

Evidence of a chemical change:

The block contains four photographs, each labeled with a letter (a, b, c, d) and a description of a chemical change.
a. A beaker containing a white liquid with bubbles forming on the surface.
b. A beaker containing a yellow liquid with a solid yellow precipitate forming at the bottom.
c. A crucible containing a substance that is burning with a bright flame and producing a large amount of white smoke.
d. A beaker containing a colorless liquid that has turned a pink color.

a. When acetic acid, vinegar, and sodium bicarbonate, baking soda, are mixed, the solution bubbles as carbon dioxide forms.

b. When solutions of sodium sulfide and cadmium nitrate are mixed, cadmium sulfide, a solid precipitate, forms.

c. When aluminum reacts with iron(III) oxide in the clay pot, energy is released as heat and light.

d. When phenolphthalein is added to ammonia dissolved in water, a color change from colorless to pink occurs.

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