

1.1. What is the main purpose of the following experiment?

$$P_{\text{H}_2} = \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}} \times \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}} \times \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}$$

- To determine the rate of reaction
- To determine the order of reaction
- To determine the activation energy
- To determine the rate constant

1.2. What is the rate of reaction?

- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time

1.3. What is the rate of reaction?

- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time

$$P_{\text{H}_2} = \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}} \times \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}} \times \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}$$

1.4. What is the rate of reaction?

- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time

1.5. What is the rate of reaction?

- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time

$$P_{\text{H}_2} = \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}} \times \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}} \times \frac{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}{P_{\text{H}_2}^{\text{atm}} - P_{\text{H}_2}^{\text{vap}}}$$

1.6. What is the rate of reaction?

- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time
- The change in concentration of a reactant or product per unit time

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Joseph William Mellor



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