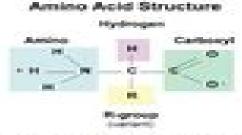
#### Study Guide: Biochemistry exam 1

- Linetteen 2:
  - Define or describe the "Central Dagma" in relation to the storage & retrieval of genetic information, the creation of proteins and the expression of traits:
    - "Central Dogma" explains the storage, retrieval, & expression of genetic information.
    - Replication: displication of DNA to make an identical copy.
    - Transcription: reading of a gene & its transcription into RNA.
    - Translation: translation of RNA sequence into the corresponding sequence of amino acids to form a protein.
    - "Information processing": Information flow from DNA to protein structure
      - DNA undergoes replication → transcribed into RNA → RNA translated into a protein → proteins develop into traits.
  - Illustrate and/or describe the general structural, physical and/or chemical characteristics of amino acids.
    - There are 29 standard amino acids.
    - Each amino acid has an amino aroup, and a corboxyl aroup joined at the alpha carbon.
    - The 'R' group is the unique feature for each amino acid.



- There's an O- on the corbosy's group by's the H+ goes to amino group, making NW+ at a certain pW.
- ◆ There are 4 main groups of omino acids:
  - Neutral non-polar; no chorpe, hydrophobic;
  - Neutral polars no charge, hydrophilic
  - Acidics extra carbonyl in R group, can be charged
  - Basic: extra amino in R group, can be + charged.

Charges occur in R gresson.

- Amphoteric: when an amino acid can act as an acid or base (depending on the pH).
- Amino acids are "switterions" meaning they can carry both a + and charge on the same molecule.
- The alpha corbon is "asymmetric" if all 4 attached groups are different.
- Describe and/or illustrate the titration of an amino ocid, including the pi's or pKo's of ionizable arrays.
  - Begin titration by adding a base, and watch the pH change.
  - You can increase the pill by adding a base, thus giving CDOII the time to give up its protos.
  - pilla reflects the dissociation of an acid.
  - pl is a point where 100% of the molocule has a 0 charge.

# **Essential Biochemistry Study Guide**

**L** Reisser

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