

Isotope Practice Worksheet

Name: _____

Learning Target: Use isotope notation to determine: element name/symbol, atomic number, number of electrons, number of neutrons, number of protons, mass number, atomic number, atomic mass.
Isotope Notation:

1. Here are three isotopes of an element: ${}_6^{12}\text{C}$ ${}_6^{13}\text{C}$ ${}_6^{14}\text{C}$
 - a. The element is: _____carbon_____
 - b. The number 6 refers to the _____atomic number_____
 - c. The numbers 12, 13, and 14 refer to the _____atomic mass_____
 - d. How many protons and neutrons are in the first isotope? ____P=6 N=6_____
 - e. How many protons and neutrons are in the second isotope? ____P=6 N=7_____
 - f. How many protons and neutrons are in the third isotope? ____P=6 N=7_____

2. Complete the following chart:

Isotope name	atomic #	mass #	# of protons	# of neutrons	# of electrons	Isotope Notation
uranium-235	92	235	92	143	92	${}_{92}^{235}\text{U}$
uranium-238	92	238	92	146	92	${}_{92}^{238}\text{U}$
boron-10	5	10	5	5	5	${}_5^{10}\text{B}$
boron-11	5	11	5	6	5	${}_5^{11}\text{B}$

- 3.

Element	Symbol	Atomic Number	Number of electrons	Number of Neutrons	Mass number	Isotope Name	Isotope Notation
Helium	He	2	2	2	4	Helium-4	${}_2^4\text{He}$
Titanium	Ti	22	22	28	50	Titanium-50	${}_{22}^{50}\text{Ti}$
Tantalum	Ta	73	73	108	181	Tantalum-181	${}_{73}^{181}\text{Ta}$
Gallium	Ga	31	31	39	70	Gallium-70	${}_{31}^{70}\text{Ga}$
Carbon	C	6	6	7	13	Carbon-13	${}_6^{13}\text{C}$
Radium	Ra	88	88	218	226	Radium-226	${}_{88}^{226}\text{Ra}$
Bismuth	Bi	83	83	127	210	Bismuth-210	${}_{83}^{210}\text{Bi}$

Isotopes And Ions Worksheet Answers

Robert R. Redfield



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Table of Contents Isotopes And Ions Worksheet Answers

1. Understanding the eBook Isotopes And Ions Worksheet Answers
 - The Rise of Digital Reading Isotopes And Ions Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Isotopes And Ions Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Isotopes And Ions Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Isotopes And Ions Worksheet Answers
 - Personalized Recommendations
 - Isotopes And Ions Worksheet Answers User Reviews and Ratings
 - Isotopes And Ions Worksheet Answers and Bestseller Lists
5. Accessing Isotopes And Ions Worksheet Answers Free and Paid eBooks
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- Isotopes And Ions Worksheet Answers Budget-Friendly Options
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 - Isotopes And Ions Worksheet Answers Compatibility with Devices
 - Isotopes And Ions Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Isotopes And Ions Worksheet Answers
 - Highlighting and Note-Taking Isotopes And Ions Worksheet Answers
 - Interactive Elements Isotopes And Ions Worksheet Answers
- 8. Staying Engaged with Isotopes And Ions Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Isotopes And Ions Worksheet Answers
- 9. Balancing eBooks and Physical Books Isotopes And Ions Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Isotopes And Ions Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Isotopes And Ions Worksheet Answers
 - Setting Reading Goals Isotopes And Ions Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Isotopes And Ions Worksheet Answers
 - Fact-Checking eBook Content of Isotopes And Ions Worksheet Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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