

PhET Isotopes and Atomic Mass

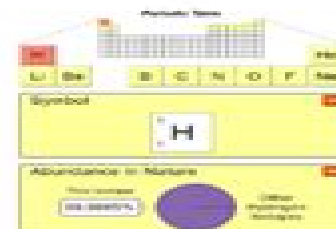
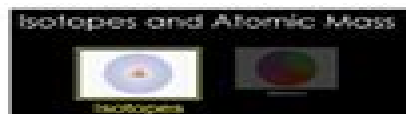
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Use the link <https://phet.colorado.edu/en/simulation/isotopes-and-atomic-mass> to go to the PhET website for the simulation Isotopes and Atomic Mass.

Begin with the **Isotopes and Atomic Mass**.

Click on the + symbols next to the bar with Symbol and Abundance in Nature to display the data.

You can build isotopes of different atoms by clicking on the symbol and adding or subtracting neutrons using the bowl of neutrons next to the scale. You can also click on the scale to change to Atomic Mass (amu).



Let's practice with Boron.

1) What are the symbol, atomic mass and abundance in nature of the isotope Boron - 11?



Now add one neutron to make Boron -12. Notice that the atom shakes. This indicates that the isotope is unstable. Unstable isotopes undergo radioactive decay. You will need to note this for other elements in the table below.

Symbol			$^{18}_8\text{O}$			
Name	Carbon-13			Beryllium-10		
Protons		7			2	
Neutrons		6			2	10
Electrons						
Atomic Mass						18.99840
Abundance						
Stable OR Unstable						

Isotopes And Atomic Mass Phet Lab Answers

Richard B. Firestone



Isotopes And Atomic Mass Phet Lab Answers:

Government Reports Announcements & Index, 1992 **Structure of the Atom, Relative Atomic Mass & Isotopes**
,2006 *The Masses of Heavy Isotopes* Martin O. Stern,1948 Table of Isotopes Richard B. Firestone,1999-08-03 Nuclear structure and decay data for thousands of isotopes and isomers a new update of the definitive reference The 1999 Table of Isotopes booklet set features Nuclear structure and radioactive decay data for approximately 3 700 isotopes and isomers an increase of more than 100 since the 1998 Update Up to date mass chain information with more than 10% revised material The latest versions as of December 1998 of the Evaluated Nuclear Structure Data File ENSDF and Nuclear Science Reference NSR file Additional data from several evaluation sources including The Table of Superdeformed Nuclear Bands and Fission Isomers Updated isotope summary table as well as energy ordered gamma ray and alpha particle tables now included on the software available from the book s website at www.wiley-vch.de/books/info/0_471_35633_6 Updated appendices for elemental data nuclear charts and gamma ray energy standards Adoption of the 1997 IUPAC recommended heavy element names More than 25 000 references Convenient links to additional atomic mass nuclear astrophysics rates spontaneous fission thermal neutron capture and more Plus the Isotope Explorer 2 22 software lets you search the entire database by level scheme drawings annotated tables data plots nuclear structure charts and keywords as well as download the latest data directly from the Table of Isotopes Web site The 1999 Update booklet features a new comprehensive isotope nuclear structure table In addition it provides clear step by step instructions on navigating the Table of Isotopes electronic content and accessing its Web site System Requirements PC Macintosh r or UNIX r systems with double speed and sufficient RAM to run Adobe TM Acrobat r see Adobe Acrobat Reader information in the book for specific system requirements Included Adobe Acrobat Reader r 3 02 for Windows r 95 98 and NT and Linux 1 2 13 or higher Acrobat Reader 3 01 for Windows 3 1 Isotope Explorer 2 22 for Windows 95 98 and NT **Atomic Mass Determinations for Some Isotopes of Ti, W and Hg** Kenneth Stephen Kozier,2012 Atomic Mass Determinations for Some Isotopes of Ti, W and Hg Kozier, Kenneth Stephen,1977 **Atomic Masses of the Isotopes of Iron, Cobalt, Nickel, Copper, and Zinc** Karl Spangler Quisenberry,1955 *Atomic Mass Differences for Some Rare Earth Isotopes Near N* Roy Lovitt Bishop,2012 **Precise Atomic Mass Differences Amongst Isotopes of Hg, Tl, Pb, and Bi** Vladimir Peter Derenchuk,1980 Atomic Mass Differences for Some Rare Earth Isotopes Near N Bishop, Roy Lovitt,1969

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