

## Predicting Products of Chemical Reactions

This worksheet is designed to help you predict products of simple reactions of the four basic reaction types (synthesis, decomposition, single replacement, and double replacement) and combustion reactions.

For the first few reactions, the type of reaction is listed, you should predict the products, then balance. Further questions just have the reactants listed and you should decide on the type of reaction, as well as the correct products. Many of these reactions fall into the category of redox reactions, though do not let that confuse you...each can be described in terms of the four basic reaction types (except the combustion reactions).

Although states (s, l, g, aq) of the reactants and products are very important in a chemical reaction, don't worry about determining those for these problems. Rather, focus on what products might result from the reactants given. Pay particular attention to the ionic charge of species that you know form ions with only one possible charge (e.g., alkali metals, alkaline earth metals, halogens, etc.)

- |                         |                |   |            |               |
|-------------------------|----------------|---|------------|---------------|
| a. Combustion:          | $C_4H_{12}$    | + | $O_2$      | $\rightarrow$ |
| b. Combustion:          | $C_4H_8$       | + | $O_2$      | $\rightarrow$ |
| c. Combustion:          | $C_4H_{10}O_2$ | + | $O_2$      | $\rightarrow$ |
| 1. Synthesis:           | $Mg$           | + | $I_2$      | $\rightarrow$ |
| 2. Double displacement: | $CuCl_2$       | + | $H_2S$     | $\rightarrow$ |
| 3. Double displacement: | $NaOH$         | + | $HClO_4$   | $\rightarrow$ |
| 4. Decomposition:       | $ZnCO_3$       | + | heat       | $\rightarrow$ |
| 5. Single replacement:  | $HCl$          | + | $Zn$       | $\rightarrow$ |
| 6. _____                | $Na$           | + | $MgCl_2$   | $\rightarrow$ |
| 7. _____                | $CaCl_2$       | + | $K_2CO_3$  | $\rightarrow$ |
| 8. _____                | $K$            | + | $Cl_2$     | $\rightarrow$ |
| 9. _____                | $BaCl_2$       | + | $K_3PO_4$  | $\rightarrow$ |
| 10. _____               | $H_2SO_4$      | + | $KOH$      | $\rightarrow$ |
| 11. _____               | $Al_2(CO_3)_3$ | + | heat       | $\rightarrow$ |
| 12. _____               | $Al$           | + | $O_2$      | $\rightarrow$ |
| 13. _____               | $Pb(NO_3)_2$   | + | $KOH$      | $\rightarrow$ |
| 14. _____               | $H_2SO_4$      | + | $BaCl_2$   | $\rightarrow$ |
| 15. _____               | $Ca$           | + | $AgCl$     | $\rightarrow$ |
| 16. _____               | $H_3PO_4$      | + | $FeBr_3$   | $\rightarrow$ |
| 17. _____               | $Li$           | + | $N_2$      | $\rightarrow$ |
| 18. _____               | $HCl$          | + | $Mg(OH)_2$ | $\rightarrow$ |
| 19. _____               | $Mg(OH)_2$     | + | heat       | $\rightarrow$ |
| 20. _____               | $Fe(OH)_3$     | + | heat       | $\rightarrow$ |

# If8766 Predicting Products Of Chemical Reactions Answers

**Liyong Dong**



## **If8766 Predicting Products Of Chemical Reactions Answers:**

## **If8766 Predicting Products Of Chemical Reactions Answers** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **If8766 Predicting Products Of Chemical Reactions Answers**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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