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already have full octets so the increasing trend stops at the halogens and the noble gases have extremely low values the
element with the highest electron affinity is fluorine

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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i
1 900 kJ mol i 2 1757 kJ mol which of the following identifies the most probable value for the third ionization energy for be and provides the best justification

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