



MERCY JUNIOR COLLEGE
Grade 11
Physical Science
Finals Exam

NAME: _____

YEAR LEVEL & SECTION: _____

TEACHER'S NAME: _____

PROCTOR'S NAME: _____

1. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol e^- or β^- , with a negative elementary electric charge.
A. Neutron
B. Electron
C. Proton
D. Atom
- It is a subatomic particle, symbol p or p^+ , with a positive electric charge of $+1e$ elementary charge and mass slightly less than that of a neutron.
A. Neutron
B. Electron
C. Proton
D. Atom
- It is a subatomic particle, symbol n or n^0 , with no net electric charge and a mass slightly larger than that of a proton.
A. Neutron
B. Electron
C. Proton
D. Atom
- Is this an example of ionic bonding?



carbon dioxide

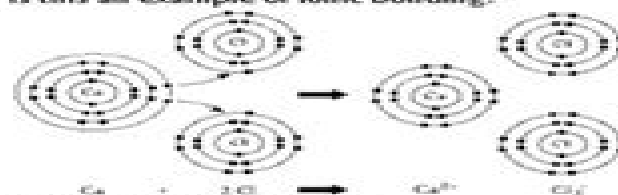
- Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.

- Is this an example of ionic bonding?



lithium fluoride

- Yes, it is ionic bonding.
 - No, it is of another type of chemical bonding.
 - There is no chemical bonding present.
 - This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
A. Electron Configuration
B. Periodic Table of Elements
C. Orbital Models
D. Energy Levels
 - Is this an example of ionic bonding?



- Yes, it is ionic bonding.
 - No, it is of another type of chemical bonding.
 - There is no chemical bonding present.
 - This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
A. Octet Rule
B. Valence Electrons
C. Atomic Orbitals
D. Chemical Bonds
 - It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
A. Neutron
B. Electron
C. Proton
D. Atom
 - It is a type of chemical bond that involves the electrostatic attraction between oppositely

Grade 11 Physical Science Final Exam Limpopo

Puja Mehta



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