Name Rocio Castro

Period:

Chapter 3: Water and the Fitness of the Environment

Concept 3.1 The polarity of water molecules results in hydrogen bonding

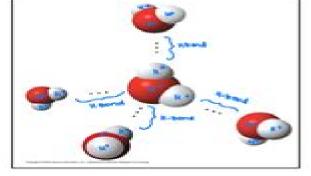
- Study the water molecules at the right. On the central molecule, label oxygen (O) and hydrogen (H).
- What is a polar molecule? Why is water considered polar? greated has it is one O Polar materials

or electrons -2 +1 fee - Overgo service by Zanton Bartoni

- Now, add + and signs to indicate the charged regions of each molecule. Then, indicate the hydrogen bonds.
- Explain Invilragen bonding. How many hydrogen bonds can a single water molecule form?

4 H-bonds may per water melecute.

is bonding: noncovered attraction between hydrogen & our electro registive, otom, mean by



Concept 3.2 Four emergent properties of water contribute to Earth's fitness for life

Hydrogen bonding accounts for the unique properties of water. Let's look at several.

Cohesion

Distinguish between cohesion and adhesion.

Constant the 8-band tention between union metales. (interminator forces)

Administrated when you see beads of water on a waxed car hood?

& surface tension

7. Which property explains the ability of a water strider to walk on water? Star Case Sevention

Moderation of Temperature

The calorie is a unit of heat. Define calorie.

The armout of heat it takes to vaise the temp of 19 of 9. Water has high specific heat. What does this mean? How does water's specific heat compare to alcohol's? Specific heat is the amount of heat that causes ig of water to change temp by 1°C. water specific heat is I early . Alcohol requires less heat to change temp since it's specific heat is 0.0%. Alcohol requires less heat 10. Explain how hydrogen bonding contributes to water's high specific heat.

Hudrogen bonds are all over water molecules & it takes a lot of heat to break the 4-80nds and a good amount of heat is released when H. Bonds form - 5 -

Because H-bonds, water molecules move faster

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