
Firefighting Exam 1 Study Questions and Answers Top Graded 2024

1. What is fire? - Fire is a rapid, self sustaining oxidation process accompanied by the evolution of heat and light in varying intensities
2. The fire triangle requires three things, what are these things? - Fuel, air, and heat
3. If you add a chemical chain reaction to the fire triangle it becomes what? - It becomes a fire tetrahedron
4. The two most common elements in fuels are? - Carbon and Hydrogen
5. Heavy, dense, black smoke being emitted by fire is also known as? - Black fire
6. What are the states of matter? - Solid, liquid, and gas
7. What is the process that converts solids and liquids to the gaseous state by adding energy? - Pyrolysis
8. What term describes when a fuel combusts and is self-sustaining? - Ignition temperature
9. What term describes the ability of a substance to mix with water? - Miscibility
10. What term refers to the weight of a liquid as compared to an equal volume of water? - Specific gravity
11. What term compares the density of a vapor to that of air? - Vapor density
12. When a liquid gives off vapors to make an ignitable mixture but will not sustain combustion is called? - Flash point
13. What is known as a mixture that the maximum concentration of gas or vapor for a mixture to ignite? - Upper flammable limit
14. What is known as a mixture that is the minimum concentration for a gas or vapor for a mixture to ignite? - Lower flammable limit
15. The mixture of a gas or vapor that will ignite between the minimum and maximum concentration is called the.....? - Flammable range
16. What method of heat transfer has a medium without visible motion? - Conduction

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N Colangelo

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web the six NO_3^-aq ions and the six Na^+aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+}\text{aq} + 2\text{PO}_4^{3-}\text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water H_2O the spectator ions in KOH and HNO_3 are K^+ and NO_3^- how to identify and list the

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web the Ag^+ and Cl^- ions become $\text{AgCl}(\text{s})$ but the Na^+ ions and the NO_3^- ions stay as Na^+ ions and NO_3^- ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be $2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} \rightarrow \text{CuCl}_2 \text{s}$ $2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} \rightarrow \text{CuCl}_2 \text{s}$ these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length