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web video answer this is a question that we are given there exists an solution with a small amount of liquid hi odeen i have

the 10 ml 0.500 M HCl off zero point is 2.02 M off acetone and 15 ml off let us assume this a one concentration of x and narcotic

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web jan 10 2023 introduction in this lab we will apply the concepts of reaction rate order of reaction and catalyst our example reaction is the iodination of acetone catalyzed by hydrochloric acid the reaction is zero order in iodine you will figure out the order in acetone and the order in HCl and the rate constant

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web video answer there can a 6.0 M solution in sodium hydroxide ensure it has been reduced to 400 liters so we have to find the moles in the solution the first thing

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web the iodination of acetone is convenient because the color of the iodine reactant is readily observable and the change in concentration of the iodine can be easily monitored by our spectrophotometer and the reaction proceeds quickly enough for multiple trials but slowly enough to obtain measurements

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web i the iodination of acetone the rate of this reaction is found to depend on the concentration of hydrogen ion in the solution as well as presumably on the concentrations of the two reactants

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web video answer the data was given by an point we arrange the dates in ascending order this is the pointing we need to build a stem and plot graph to present quantity data included a graphical input this is the stem part and this is the leaf item yours is

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