

CBSE NCERT Solutions for Class 11 Chemistry Chapter 12

Back of Chapter Questions

1. What are hybridisation states of each carbon atom in the following compounds?



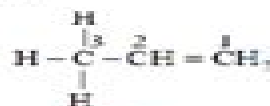
Solution:

Hybridisation = σ bond + lone pairIf $2\sigma = sp$ $3\sigma = sp^2$ $4\sigma = sp^3$

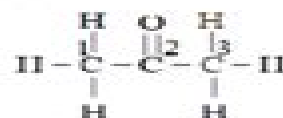
(i)

C - 1 is sp^2 hybridisedC - 2 is sp hybridized

(ii)

C - 3 is sp^3 hybridisedC - 2 is sp^2 hybridisedC - 1 is sp^2 hybridised

(iii)

C - 1 and C - 3 are sp^3 hybridised

Modern Chemistry Ch 12 Tests A And

Michael Brown



Modern Chemistry Ch 12 Tests A And:

Reviewing **Modern Chemistry Ch 12 Tests A And**: Unlocking the Spellbinding Force of Linguistics

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