#### **Nuclear Reactions**

Nuclear reactions are characterized by the following:

- They involve changes to the nucleus
  - Number of protons or neutrons in the nucleus change
- Energy changes are enormous
- · Not concerned with valence electrons.
- Changes in temperature and pressure have no effect on the rate of the reaction.
- Small particles are important.

#### Important subatomic particles

proton: P H

neutron: !n

Migha Particle: 4 0x 4 He anuvirons

beta particle: "B" e - high energy elections emitted from the nucleus position: "B" fe - positive elections what gives a proton its positive charge

Gamma Ray: "Y - high energy proton

#### Definitions:

Isotopes – same number of protons but different mass number (mass number = # protons + # neutrons)

16C 15C are each an isotope of C (How many neutrons does each have?)

Nuclide (pronounced NOO clyde) - a nucleus with a specific number of protons and neutrons

Radioactive - Changes in matter resulting from SPONTANEOUS changes in the composition of the nucleus.

- Such changes are often accompanied by the creation of the subatomic particles above.
- Changes to the nucleus often result in the conversion of one element into another.

Radionuclide - nuclides that are radioactive.

"C — "N " "C p particle-trigh energy or emisted from moves."

236 U — 200 Th · 4H

Talso 4 or -alpea particle

# **Modern Chemistry 22review Nuclear Chemistry Answers**

**R Pring** 

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