

Ions and Isotopes Worksheet

Symbol	Name of the element	Atomic number	Protons	Neutrons	Electrons	Atomic Mass	Charge
O -17	Oxygen	8	8	9	8	17	0
Xe - 131	Xenon	54	54	77	54	131	0
$^{207}_{82}\text{Pb}$	Lead	82	82	125	82	207	0
Na¹⁺	Sodium ion	11	11	12	10	23	+1
Sr	Strontium	38	38	50	38	88	0
Al³⁺	Aluminum ion	13	13	14	10	27	+3
$^{238}_{92}\text{U}$	Uranium	92	92	146	92	238	0
$^{75}_{33}\text{As}^{-3}$	Arsenic	33	33	39	36	75	-3
$^{32}_{16}\text{S}^{-2}$	Sulfide ion	16	16	16	18	32	-2
Na	Sodium	11	11	12	11	23	0
N³⁻	Nitride ion	7	7	7	10	14	-3
$^{65}_{29}\text{Cu}$	Copper	29	29	36	29	65	0
$^{120}_{80}\text{Hg}$	Mercury isotope	80	80	40	80	120	0
Ag	silver	47	47	61	47	108	0
Zr	zirconium	40	40	51	91	40	0
Cr	chromium	24	24	28	24	56	0
$^{261}_{104}\text{Rf}$	rutherfordium	104	104	158	104	261	0

Isotopes And Ions Worksheet Answer Key

Minjie Lin

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Ion vs. Isotope | Understanding Ions, Isotopes and Properties of Elements | Grade 6-8 Physical Science Dot
EDU,2024-04-15 Unlock the mysteries of the atomic world with Ion vs Isotope Understanding Ions Isotopes and Properties of Elements Tailored for grades 6 8 this book demystifies the complex concepts of ions and isotopes and their crucial roles in the composition of matter Essential for educators homeschooling parents and school librarians it supports the US STEM curriculum making it a perfect resource for young scientists eager to explore the building blocks of everything in the universe

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