



## EMCP II+ OP5-0 setpoints

P001	Fuel solenoid type	0=ETR, 1=ETS
P002	Units shown	0=Eng, metric
P003	Shutdown override for Engine fault	0=Shutdown, 1=Alarm
P004	Shutdown override for for Sensor fault	0=Alarm, 1=Shutdown
P005	Coolant loss sensor installed	0=w/o sensor, 1=w/sensor
P006	Shutdown override for coolant loss fault	0= shutdown, 1= alarm
P007	System voltage	24 volts or 32 volts
P008	Not used	NA
P009	Number of ring gear teeth	95 to 350 teeth
P010	Engine overspeed	500 to 4330 rpm
P011	Crank terminate speed	100 to 1000 rpm
P012	Oil stop speed	400 to 1800 rpm
P013	Low oil pressure shutdown at Rated speed	34 to 240 kPa (5 to 61 psi)
P014	Low oil pressure shutdown at Idle speed	20 to 336 kPa (3 to 49 psi)
P015	High water temperature shutdown	85 to 123 C (185° to 253°F)
P016	Low water temperature alarm	0 to 36°C (32 to 97°F)
P017	Total cycle crank time	5 to 350 seconds
P018	Cycle crank time	5 to 300 seconds
P019	Cooldown time	0 to 30 minutes
P020	AC voltage	150V to 30.0 kV
P021	AC current full scale	75A to 4000A
P022	GSC Engine Number	01 to 99
P023	Engine type	0=MUI, 1=Gas, 2=EUI
P024	Crank time delay (only for gas)	0 to 20 seconds
P025	Oil temperature sensor installed	0=w/o sensor, 1=w/sensor
P026	High oil temperature shutdown	85 to 123°C (185 to 253°F)
P027	Shutdown enable for high oil temperature	0=alarm, 1=shutdown
P028	Name plate voltage	100 to 25kV
P029	Name plate Current	0 to 4000A
P030	Name plate power	0 to 10MW
P031	Rated frequency	50, 60 or 400 Hz
P032	Connection configuration of generator	0=wye, 1=delta
P033	Number of generator poles	0 to 254

# Emcp Ii Codes

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negative when heat flows from the system to the surroundings thus exothermic reactions have a negative  $\Delta h$  of reaction the  
word exothermic is derived from exo the greek word for outside and therme the greek word

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example  $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the  $\text{h}^+$  ions from the acid and the  $\text{oh}^-$  ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol<sup>-1</sup>

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