

Arrangement of Electrons in Atoms

Section Quiz: Electron Configurations

In the space provided, enter the letter of the atom or phrase that best completes each statement or best answers each question.

- The statement that electrons orbit the nucleus in discrete energy levels is based on
 - Bohr's theory.
 - Strom's model.
 - the quantum principle.
 - the Pauli exclusion principle.
- According to the Aufbau principle, which of the following orbitals is filled last?
 - 1s
 - 2s
 - 2p
 - 3s
- In a ground-state hydrogen atom, which orbital is the electron in?
 - 1s
 - 2s
 - 2p
 - 3s
- What is the maximum number of electrons in the outermost shell of the element with configuration $1s^2 2s^2 2p^6 3s^2 3p^4$?
 - 6
 - 10
 - 16
 - 18
- The ground-state electron configuration of bromine is $1s^2 2s^2 2p^6 3s^2 3p^4$. In this configuration, what number of electrons is contained in the completely filled?
 - 1
 - 2
 - 10
 - 18

Modern Chemistry Arrangement Of Electrons In Atoms

Section Quiz

Clemens Wendtner



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