



Chemical Reactions and Equations

Chemical reactions are the process in which new substance with new properties are formed.

The burning of magnesium in air to form magnesium oxide is an example of a chemical reaction



Note:- The magnesium ribbon which we use usually has a coating of 'magnesium oxide' on its surface which is formed by the slow action of oxygen of air on it. So before burning in air, the magnesium ribbon is cleaned by rubbing with a sand paper. This is done to remove the protective layer of magnesium oxide from the surface of magnesium ribbon so that it may readily combine with the oxygen of air.

Characteristics of Chemical Reactions

- (i) Evolution of a gas
- (ii) Formation of a precipitate
- (iii) Change in colour
- (iv) Change in temperature
- (v) Change in state

1.) Evolution of a gas

The chemical reaction b/w zinc and dilute sulphuric acid is characterized by the evolution of Hydrogen gas



Modern Chemistry Chemical Equations And Reactions Notes

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Modern Chemistry Chemical Equations And Reactions Notes:

Modern Chemistry Chemical Equations And Reactions Notes Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Modern Chemistry Chemical Equations And Reactions Notes**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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