

Name: \_\_\_\_\_

## Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

### PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

### PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

### PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

| <u>Double Replacement Reactions</u> |                                 |   |                                   |   | <b>Precipitates<br/>(if any)</b> |
|-------------------------------------|---------------------------------|---|-----------------------------------|---|----------------------------------|
| A.                                  | NaCl                            | + | AgNO <sub>3</sub>                 | → |                                  |
| B.                                  | KI                              | + | AgNO <sub>3</sub>                 | → |                                  |
| C.                                  | NaOH                            | + | AgNO <sub>3</sub>                 | → |                                  |
| D.                                  | Na <sub>3</sub> PO <sub>4</sub> | + | AgNO <sub>3</sub>                 | → |                                  |
| E.                                  | NaCl                            | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| F.                                  | KI                              | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| G.                                  | NaOH                            | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| H.                                  | Na <sub>3</sub> PO <sub>4</sub> | + | Pb(NO <sub>3</sub> ) <sub>2</sub> | → |                                  |
| I.                                  | NaCl                            | + | CuSO <sub>4</sub>                 | → |                                  |
| J.                                  | KI                              | + | CuSO <sub>4</sub>                 | → |                                  |
| K.                                  | NaOH                            | + | CuSO <sub>4</sub>                 | → |                                  |
| L.                                  | Na <sub>3</sub> PO <sub>4</sub> | + | CuSO <sub>4</sub>                 | → |                                  |

# Lab 10 2 Double Replacement Reactions

**Richard Bailey**



## **Lab 10 2 Double Replacement Reactions:**

## The Enigmatic Realm of **Lab 10 2 Double Replacement Reactions**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Lab 10 2 Double Replacement Reactions** a literary masterpiece penned by a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those that partake in its reading experience.

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