

Ionic and Covalent Bonds Study Guide

Be able to define the following terms:

- | | | |
|-------------------|-----------------|--------------------|
| ➤ covalent bond | ➤ anion | ➤ polyatomic ion |
| ➤ Lewis structure | ➤ cation | ➤ hydrogen bonding |
| ➤ molecule | ➤ chemical bond | ➤ valence electron |
| ➤ polar bond | ➤ ionic bond | |
| ➤ non-polar bond | ➤ ion | |

1. A group of covalently bonded atoms that acts together as one charged atom is a polyatomic ion.
2. A chemical bond that occurs when atoms share electrons is a(n) covalent bond.
3. The octet rule tells you how many electrons an atom must gain, lose, or share to become stable.
4. The oxidation number of an atom is shown with a roman numeral in parenthesis.
5. How many electrons are needed in the outer energy levels of most atoms for the atom to be chemically stable? **8 (Eight)**
6. Why do the noble gases NOT form compounds readily? **They have 8 electrons in their outer electron shell.**
7. What is the total number of atoms in the compound $\text{Ca}(\text{ClO}_3)_2$? **9 (Nine)**
8. Explain why it is correct to say that a compound like calcium carbonate, CaCO_3 , contains both ionic and covalent bonds. **Carbonate is a polyatomic ion (covalent bonded) and then it forms an ionic bond with Calcium.**
9. Name or write the formulas for the following ionic compounds:
 - a. Al_2O_3 **Aluminum Oxide**
 - b. NH_4Cl **Ammonium Chloride**
 - c. FeO **Iron (II) Oxide**
 - d. manganese (V) oxide **Mn_2O_5**
 - e. calcium chloride **CaCl_2**
 - f. lead (IV) nitrate **$\text{Pb}(\text{NO}_3)_4$**

Ionic And Covalent Bonds Study Guide Answers

Judd E. Hollander



Ionic And Covalent Bonds Study Guide Answers:

This book delves into Ionic And Covalent Bonds Study Guide Answers. Ionic And Covalent Bonds Study Guide Answers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Ionic And Covalent Bonds Study Guide Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Ionic And Covalent Bonds Study Guide Answers
- Chapter 2: Essential Elements of Ionic And Covalent Bonds Study Guide Answers
- Chapter 3: Ionic And Covalent Bonds Study Guide Answers in Everyday Life
- Chapter 4: Ionic And Covalent Bonds Study Guide Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Ionic And Covalent Bonds Study Guide Answers. The first chapter will explore what Ionic And Covalent Bonds Study Guide Answers is, why Ionic And Covalent Bonds Study Guide Answers is vital, and how to effectively learn about Ionic And Covalent Bonds Study Guide Answers.
3. In chapter 2, the author will delve into the foundational concepts of Ionic And Covalent Bonds Study Guide Answers. The second chapter will elucidate the essential principles that must be understood to grasp Ionic And Covalent Bonds Study Guide Answers in its entirety.
4. In chapter 3, this book will examine the practical applications of Ionic And Covalent Bonds Study Guide Answers in daily life. The third chapter will showcase real-world examples of how Ionic And Covalent Bonds Study Guide Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Ionic And Covalent Bonds Study Guide Answers in specific contexts. The fourth chapter will explore how Ionic And Covalent Bonds Study Guide Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Ionic And Covalent Bonds Study Guide Answers. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Ionic And Covalent Bonds Study Guide Answers.

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