

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. C In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
(a) how many atoms of Fe are in one formula unit
(b) how many sulfate ions can be attached to the iron atom
(c) the charge on each Fe ion
(d) the total positive charge of the formula unit
2. C The result of changing a subscript in a correctly written chemical formula is to _____
(a) change the number of moles represented by the formula
(b) change the charges on the other ions in the compound
(c) change the formula so that it no longer represents the compound it previously represented
(d) have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
4 elements a. How many elements make up this compound?
6 oxygen atoms b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
21 atoms c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
11 atoms a. $Ca(HCO_3)_2$
45 atoms b. $C_{12}H_{22}O_{11}$
10 atoms c. $Fe(ClO_2)_3$
9 atoms d. $Fe(ClO_3)_2$
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_3 c. What is the formula for sulfurous acid?
phosphoric acid d. What is the name for the acid H_3PO_4 ?

Ionic Bonds Chapter 7 Section 7 2 Answers

Sabine Zange



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Table of Contents Ionic Bonds Chapter 7 Section 7 2 Answers

1. Understanding the eBook Ionic Bonds Chapter 7 Section 7 2 Answers
 - The Rise of Digital Reading Ionic Bonds Chapter 7 Section 7 2 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bonds Chapter 7 Section 7 2 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Bonds Chapter 7 Section 7 2 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bonds Chapter 7 Section 7 2 Answers
 - Personalized Recommendations
 - Ionic Bonds Chapter 7 Section 7 2 Answers User Reviews and Ratings
 - Ionic Bonds Chapter 7 Section 7 2 Answers and Bestseller Lists
5. Accessing Ionic Bonds Chapter 7 Section 7 2 Answers Free and Paid eBooks
 - Ionic Bonds Chapter 7 Section 7 2 Answers Public Domain eBooks

- Ionic Bonds Chapter 7 Section 7 2 Answers eBook Subscription Services
- Ionic Bonds Chapter 7 Section 7 2 Answers Budget-Friendly Options
- 6. Navigating Ionic Bonds Chapter 7 Section 7 2 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Bonds Chapter 7 Section 7 2 Answers Compatibility with Devices
 - Ionic Bonds Chapter 7 Section 7 2 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Bonds Chapter 7 Section 7 2 Answers
 - Highlighting and Note-Taking Ionic Bonds Chapter 7 Section 7 2 Answers
 - Interactive Elements Ionic Bonds Chapter 7 Section 7 2 Answers
- 8. Staying Engaged with Ionic Bonds Chapter 7 Section 7 2 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Bonds Chapter 7 Section 7 2 Answers
- 9. Balancing eBooks and Physical Books Ionic Bonds Chapter 7 Section 7 2 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Bonds Chapter 7 Section 7 2 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Bonds Chapter 7 Section 7 2 Answers
 - Setting Reading Goals Ionic Bonds Chapter 7 Section 7 2 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Bonds Chapter 7 Section 7 2 Answers
 - Fact-Checking eBook Content of Ionic Bonds Chapter 7 Section 7 2 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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