

Chemistry: Ionic & Covalent Bonds Unit Study Guide

Covalent Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Covalent Compounds.

- Bondable
- Large / Small
- Atoms / Ions
- High / Low
- Conducts / Does not Conduct Electricity
- Solid / Liquid / Gas at Room Temperature

2. Write the prefix following number.

Number	Prefix
1	
2	
3	
4	
5	

3. What are the two

4. Complete the chart

Formula	Name
SO_2	
	Carbon
H_2O	

5. What does VSEPR

Name: _____ Date: _____ Period: _____

Ionic Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Ionic Compounds.

- Bond between a Metal / Nonmetal and a Metal / Nonmetal
- Large / Small Electronegativity Difference
- The (Metal / Nonmetal) form Cations which (Gain / Lose) Electrons
- The (Metal / Nonmetal) form Anions which (Gain / Lose) Electrons
- High / Low Melting & Boiling Point
- Conducts / Does not Conduct Electricity in Solution
- Solid / Liquid / Gas at Room Temperature

2. What charge would the following periodic table families have in a chemical bond?

Family	Charge
Alkali Metals	
Alkaline Earths	
Transition Metals	
Carbon	
Nitrogen	
Halogens	

3. Write the roman numeral that is represented by the following numbers.

Roman Numeral	Number
	1
	2
	3
	4
	5
	6

4. Complete the chart below by filling in the missing names.

Cation	Anion	Formula	Name
Mg^{2+}	Br^-		
		CuO	
			Iron (II) chloride
		$(NH_4)_3PO_4$	

Ionic And Covalent Bonds Study Guide

**Barron's Educational Series, Gabrielle
I. Edwards, Cynthia Pfirrmann**



Ionic And Covalent Bonds Study Guide:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Ionic And Covalent Bonds Study Guide**

In a global inundated with displays and the cacophony of quick interaction, the profound power and mental resonance of verbal beauty often fade in to obscurity, eclipsed by the constant assault of noise and distractions. However, set within the lyrical pages of **Ionic And Covalent Bonds Study Guide**, a captivating perform of fictional beauty that impulses with fresh thoughts, lies an wonderful trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that interesting opus instructions visitors on a psychological odyssey, gently revealing the latent potential and profound impact stuck within the elaborate web of language. Within the heart-wrenching expanse with this evocative analysis, we can embark upon an introspective exploration of the book is main styles, dissect its interesting writing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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