

Acids and Bases

MERCK

Acid				Base			
Strength of Acid	Ka	Name	Formula	Formula	Name	Strength of Base	
Strongest	Large	Perchloric acid	HClO ₄	ClO ₄ ⁻	Perchlorate ion	Weakest	
	3.2×10 ⁸	Hydroiodic acid		I ⁻	Iodide		
	1.0×10 ⁹	Hydrobromic acid	HBr	Br ⁻	Bromide		
	1.3×10 ⁴	Hydrochloric acid	HCl	Cl ⁻	Chloride		
	1.0×10 ³	Sulfuric acid	H ₂ SO ₄	HSO ₄ ⁻	Hydrogen sulfate ion		
	2.4×10 ¹	Nitric acid	HNO ₃	NO ₃ ⁻	Nitrate ion		
		Hydronium ion	H ₃ O ⁺	H ₂ O	Water		
	5.4×10 ⁻²	Oxalic acid	HO ₂ C ₂ O ₂ H	HO ₂ C ₂ O ₂ ⁻	Hydrogen oxalate ion		
	1.3×10 ⁻²	Sulfurous acid	H ₂ SO ₃	HSO ₃ ⁻	Hydrogen sulfite ion		
	1.0×10 ⁻²	Hydrogen sulfate ion	HSO ₄ ⁻	SO ₄ ²⁻	Sulfate ion		
	7.1×10 ⁻³	Phosphoric acid	H ₃ PO ₄	H ₂ PO ₄ ⁻	Dihydrogen phosphate ion		
	7.2×10 ⁻⁴	Nitrous acid	HNO ₂	NO ₂ ⁻	Nitrite ion		
	6.6×10 ⁻⁴	Hydrofluoric acid	HF	F ⁻	Fluoride ion		
	1.8×10 ⁻⁴	Methanoic acid	HCO ₂ H	HCO ₂ ⁻	Methanoate ion		
	6.3×10 ⁻⁵	Benzoic acid	C ₆ H ₅ COOH	C ₆ H ₅ COO ⁻	Benzoate ion		
	5.4×10 ⁻⁵	Hydrogen oxalate ion	HO ₂ C ₂ O ₂ ⁻	O ₂ C ₂ O ₂ ²⁻	Oxalate ion		
	1.8×10 ⁻⁵	Ethanoic acid	CH ₃ COOH	CH ₃ COO ⁻	Ethanoate (acetate) ion		
	4.4×10 ⁻⁷	Carbonic acid	H ₂ CO ₃	HCO ₃ ⁻	Hydrogen carbonate ion		
	1.1×10 ⁻⁷	Hydrosulfuric acid	H ₂ S	HS ⁻	Hydrogen sulfide ion		
	6.3×10 ⁻⁸	Dihydrogen phosphate ion	H ₂ PO ₄ ⁻	HPO ₄ ²⁻	Hydrogen phosphate ion		
	6.2×10 ⁻⁸	Hydrogen sulfite ion	HS ⁻	SO ₃ ²⁻	Sulfite ion		
	2.9×10 ⁻⁸	Hypochlorous acid	HClO	ClO ⁻	Hypochlorite ion		
	6.2×10 ⁻¹⁰	Hydrocyanic acid	HCN	CN ⁻	Cyanide ion		
	5.8×10 ⁻¹⁰	Ammonium ion	NH ₄ ⁺	NH ₃	Ammonia		
	5.8×10 ⁻¹⁰	Boric acid	H ₃ BO ₃	H ₂ BO ₃ ⁻	Dihydrogen borate		
	4.7×10 ⁻¹¹	Hydrogen carbonate ion	HCO ₃ ⁻	CO ₃ ²⁻	Carbonate ion		
	4.2×10 ⁻¹³	Hydrogen phosphate ion	HPO ₄ ²⁻	PO ₄ ³⁻	Phosphate ion		
	1.8×10 ⁻¹³	Dihydrogen borate ion	H ₂ BO ₃ ⁻	HBO ₃ ²⁻	Hydrogen borate ion		
	1.3×10 ⁻¹³	Hydrogen sulfide ion	HS ⁻	S ²⁻	Sulfide ion		
	1.6×10 ⁻¹⁴	Hydrogen borate ion	HBO ₃ ²⁻	BO ₃ ³⁻	Borate ion		
Weakest		water	H ₂ O	OH ⁻	Hydroxide	Strongest	



Lab 16 1 Creating Acid Base Strength Table

VM Jensen



Lab 16 1 Creating Acid Base Strength Table:

The Enigmatic Realm of **Lab 16 1 Creating Acid Base Strength Table**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Lab 16 1 Creating Acid Base Strength Table** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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