

### Final Study Guide

- It is called the universal solvent because of its ability to dissolve most solutes
- Water molecules are held together by hydrogen bonds

**Reflections:** Although there are homogeneities at the national level

**Results:** The authors are looking forward.

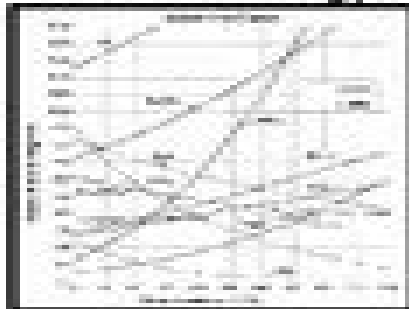
**Solvent.**—The substance dissolving the dissolving the substance there is a greater quantity

LIKNE: DISSIMILATION LIKNE- which means polar substances dissolve polar substances and nonpolar substances dissolve nonpolar substances

Soil: Soluble Silicates: Insol. ...	Clay: Insoluble Silicates: Insol. ...
At high temperatures	At low temperatures
With stirring	At high pressures
When particles are small	With, or stirring or shaking

**Colligative Properties:** Properties that depend on the concentration of a solution.

**Flow Diagrams, point diagrams, facilities, point elevations**



**Solubility curve** - a graph used to show the dependence of solubility on temperature

**Keywords:** *social capital, social networks, social capital, social capital, social capital*

**The Insurance Fund** -- "We aren't looking at any more collecting on this bond."

**Supersaturated** — very unstable solution, crystallizes easily, above the line



**Electrolyte:** A solution that conducts electricity.



**Notes—Director's Office:** A collection that does not constitute directorship. I have not discussed it here.

**Abstract**

**Abundant** — is a characteristic that has an increase of something (e.g. a large cell)

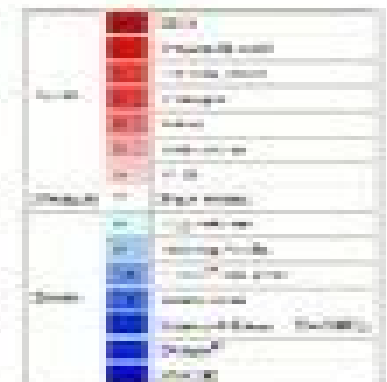
**Properties:** acid, corrosive, change red litmus paper blue, neutralizes when mixed with bases.

**Blasen** — a collection of fluid that has no memory of negative ions (CMI-5) have high pH

**Properties:** Shiny, slippery, change blue litmus paper red, conductive when mixed with acids

**gill Netter:** a series that goes from E-14, a mixture of bone-ash or bone as well as bone, in.

Don't be too broad, like positive and negative ions when dissolved in water. Things that ionize are good electrolytes.



# Ipc West Study Guide

**Yicheng Fang**



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