

Atoms and Bonding • Reading/Taking Guide**Ionic Bonds** (pp. 184–189)

This section explains how atoms become ions. It describes the electrical charge of an ionic compound and what the chemical formula of an ionic compound tells you.

Use Target Reading Skills

As you read, fill in the table to compare and contrast the properties of sodium chloride with the properties of its component ions.

| | Color | State at Room Temperature | Stability |
|-----------------------------|----------------------|---------------------------|-------------------------------|
| Sodium chloride | | | |
| Elements of sodium chloride | sodium: chlorine: | sodium: chlorine: | sodium: unstable chlorine: |

Ions (pp. 185–186)

1. An atom or group of atoms that has an electric charge is called a(n) _____.

2. What happens to an atom when it loses an electron?

3. What happens to an atom when it gains an electron?

Ionic Bonds Chapter 5 Reading Notetaking Guide Answers

David Baud



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