



IONIC AND METALLIC BONDING

Practice Problems

Do your best to solve these problems on your own.

SECTION 7.1 IONS

1. How many electrons does each element of atoms have? How many protons does each atom have? How many neutrons does it have? (Look them up.)

a. Fe

b. Li

c. Br

2. How many valence electrons does each element have? (Look them up.)

a. Fe

b. Br

c. Br

3. How many valence electrons does each atom have? (Look them up.)

a. Fe

b. Br

c. Br

d. Fe

e. Br

4. How many valence electrons does each atom have? (Look them up.) How many protons does each atom have? (Look them up.)

5. How many valence electrons does each atom have? (Look them up.) How many protons does each atom have? (Look them up.)

a. Fe

b. Br

c. Br

d. Fe

e. Br

f. Br

6. How many valence electrons does each atom have? (Look them up.)

a. Fe

b. Br

c. Fe

d. Br

7. How many valence electrons does each atom have? (Look them up.)

a. Fe

b. Br

c. Br

d. Br

8. How many valence electrons does each atom have? (Look them up.)

a. Fe

b. Br

c. Br

d. Fe

e. Br

f. Br

Ionic And Metallic Bonding Practice Problems Pearson

Guillaume Favre



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