

Activity A: Small atoms	Get the Gizmo ready: <ul style="list-style-type: none"> On the PERIODIC TABLE tab, select H (hydrogen). Select the ELECTRON CONFIGURATION tab. Click Reset. 	
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Introduction: Electrons are arranged in **orbitals**, **subshells**, and **shells**. These levels of organization are shown by the boxes of the Gizmo. Each box represents an orbital. The subshells are labeled with letters (*s*, *p*, *d*, and *f*) and the shells are labeled with numbers.

Question: How are electrons arranged in elements with atomic numbers 1 through 10?

1. Infer: Based on its atomic number, how many electrons does a hydrogen atom have? _____

2. Arrange: The **Aufbau principle** states that electrons occupy the lowest-energy orbital. Click once in the **1s** box to add an electron to the only orbital in the *s* subshell of the first shell.

Click **Check**. What is the electron configuration of hydrogen? _____

3. Arrange: Click **Next element** to select helium. Add another electron to the **1s** orbital. The arrows represent the **spin** of the electron. What do you notice about the arrows?

The **Pauli exclusion principle** states that electrons sharing an orbital have opposite spins.

4. Check your work: Click **Check**. What is the electron configuration of helium? _____

5. Arrange: Click **Next element** and create electron configurations for lithium, beryllium, and boron. Click **Check** to check your work, and then list each configuration below:

Lithium: _____ Beryllium: _____ Boron: _____

6. Arrange: Click **Next element** to select carbon. Add a second electron to the first **2p** orbital.

Click **Check**. What feedback is given? _____

7. Rearrange: **Hund's rule** states that electrons will occupy an empty orbital when it is available in that subshell. Rearrange the electrons within the **2p** subshell and click **Check**.

Is the configuration correct now? _____

Show the correct configuration in the boxes at right:

1s ☐

2s ☐ 2p ☐ ☐ ☐

(Activity A continued on next page)

Ionic Bonds Explore Learning Answer Key

N Noddings



Ionic Bonds Explore Learning Answer Key:

Whispering the Secrets of Language: An Mental Journey through **Ionic Bonds Explore Learning Answer Key**

In a digitally-driven earth where displays reign supreme and instant connection drowns out the subtleties of language, the profound techniques and psychological subtleties concealed within phrases usually go unheard. However, nestled within the pages of **Ionic Bonds Explore Learning Answer Key** a interesting fictional treasure blinking with natural feelings, lies a fantastic quest waiting to be undertaken. Penned by an experienced wordsmith, that charming opus encourages viewers on an introspective trip, lightly unraveling the veiled truths and profound affect resonating within the material of each and every word. Within the psychological depths with this touching evaluation, we can embark upon a sincere exploration of the book is primary themes, dissect its charming publishing model, and yield to the powerful resonance it evokes strong within the recesses of readers hearts.

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