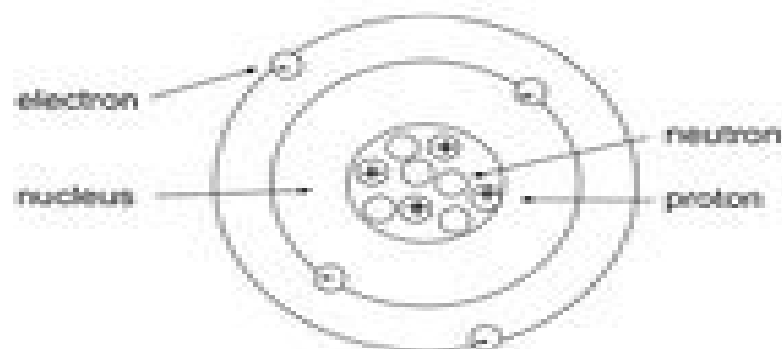


WHAT IS (the) MATTER?

Matter is anything that has mass and takes up space (volume). Matter is made up of small particles, known as atoms.

Diagram of an Atom



Atomic mass = number of _____ +
number of _____

Atomic number = number of _____

In a neutral atom,
the number of **electrons** (charge -1) is the
same as the number of **protons** (charge +1)

Problem 1: Chromium has an atomic mass of 52 and an atomic number of 24. Fill in the following:

Number of protons:

Number of neutrons:

Number of electrons:

Problem 2: Tin has an atomic mass of 118 and an atomic number of 50. Fill in the following:

Number of protons:

Number of neutrons:

Number of electrons:

Part 2: Introduction to Ions

An ion is a charged atom. How does an atom acquire a charge? When the number of electrons do not equal the number of protons, this creates a charge. Why? Remember,

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