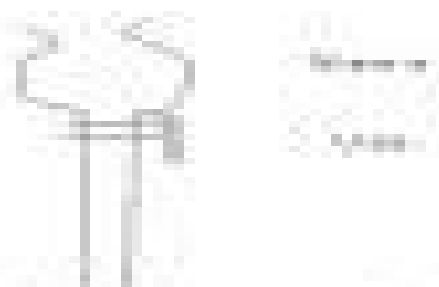


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (20%)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of H+ ions, which is used to calibrate the pH meter.

(a) Standardized solution of H+ ions is used to calibrate the pH meter.

(b) There is a reason for the choice of the pH meter.

(c) Standardized solution of H+ ions is used to calibrate the pH meter.

Form One 1 Chemistry Revision Questions

Thomas Griffiths



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