

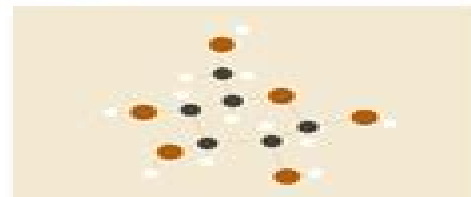
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## CHEMICAL AND PHYSICAL CHANGES

In a chemical change, a chemical reaction occurs and new products form. Energy is either absorbed or released. In a physical change, matter changes form, but its identity is not changed.

Identify each change as chemical or physical.



|    |                                                                            |  |
|----|----------------------------------------------------------------------------|--|
| 1  | Heating water into steam                                                   |  |
| 2  | Milk going sour                                                            |  |
| 3  | Dissolving sugar in water                                                  |  |
| 4  | Iron rusting                                                               |  |
| 5  | Hydrochloric acid and sodium hydroxide reacting and forming salt and water |  |
| 6  | Digesting food into nutrients                                              |  |
| 7  | Melting ice into water                                                     |  |
| 8  | Acid rain dripping onto limestone and making carbon dioxide gas            |  |
| 9  | Breaking a glass into pieces                                               |  |
| 10 | Hydrogen peroxide decomposing into oxygen gas and water                    |  |

# Matter And Change Practice Problems Answers

**Arshad Iqbal**



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