

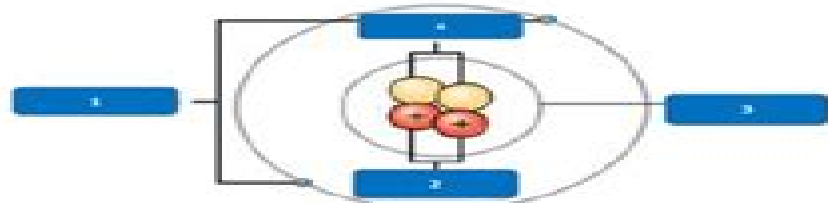


<div></div> <div>DAILY LESSON LOG</div>	School Batasan Hills National High School	Grade Level Learning Area	8 Chemistry																				
	Teacher MARIA RIZZA E. LLORENTE	Quarter 3rd																					
DLL#68	Teaching Date and Time for 50 minutes	Day: Thursday-Friday																					
Date: March 23-24, 2023																							
<b>I. OBJECTIVE:</b>																							
<b>A. Content Standard</b>	The learners demonstrate understanding of the identity of a substance according to its atomic number																						
<b>B. Performance Standard</b>	The learners should be able to present how water behaves in its different states within the water cycle.																						
<b>C. Learning Competencies</b>	Determine the number of protons, neutrons, and electrons in a particular atom; <b>(S8MT-IIIc-f-10)</b>																						
<b>Specific Objectives:</b>	At the end of the lesson, the students should be able to: 1. Describe the atom and its sub-atomic particles; 2. Determine the number of electrons, protons, and neutrons in an atom; and 3. Explain how atoms benefit life.																						
<b>II. CONTENT</b>	<b>SUBATOMIC PARTICLES</b> <b>Determining the number of protons, electrons and neutrons in an atom</b>																						
<b>III. LEARNING RESOURCES</b>																							
<b>A. References</b>																							
<b>B. Other Learning Resources</b>	<ul style="list-style-type: none"><li>• <a href="http://chemed.chem.purdue.edu/genchem/topicreview/bp/ch6/atom_emfame.html">http://chemed.chem.purdue.edu/genchem/topicreview/bp/ch6/atom_emfame.html</a></li><li>• <a href="http://www.kentchemistry.com/links/AtomicStructure/protonsws1.htm">http://www.kentchemistry.com/links/AtomicStructure/protonsws1.htm</a></li><li>• <a href="https://education2eb.org/lesson_number.html">https://education2eb.org/lesson_number.html</a></li><li>• <a href="https://www.nap.edu/read/4818/chapter/2">https://www.nap.edu/read/4818/chapter/2</a></li><li>• <a href="https://www.youtube.com/watch?v=1M4jysdgmRg">https://www.youtube.com/watch?v=1M4jysdgmRg</a></li></ul>																						
<b>IV. PROCEDURE</b>																							
<b>A. Reviewing previous lesson or presenting the new lesson.</b>	<b>ELICIT: TRIVIA QUIZ GAME</b> The class will play trivia quiz game about atoms and molecules  <a href="https://www.canva.com/design/DAP6p7UYH0Q8kUQleQA8isLMxgQK1CZueA/edit">https://www.canva.com/design/DAP6p7UYH0Q8kUQleQA8isLMxgQK1CZueA/edit</a>																						
<b>B. Establishing a purpose for the lesson</b>	<b>ENGAGE: LABEL THE PICTURE: What's Inside the Atom?</b>  <ol style="list-style-type: none"><li>a. Ask the class to label the picture.</li><li>b. Introduce subatomic particles.</li><li>c. Recall the proponent/discovery of the atomic model on the picture.</li></ol> <b>Answer: 1. Electron 2. Proton 3. Nucleus 4. neutron</b>																						
<b>C. Presenting examples/instances of the new lesson</b>	<b>ENGAGE: TABLE ANALYSIS</b> The students will study the table below and answer the questions about Properties of Subatomic Particles. <table><thead><tr><th>Subatomic particle</th><th>Symbol</th><th>Relative charge</th><th>Location</th><th>Mass</th></tr></thead><tbody><tr><td>Proton</td><td>p</td><td>+1</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Neutron</td><td>n</td><td>0</td><td>Inside the nucleus</td><td>1</td></tr><tr><td>Electron</td><td>e<sup>-</sup></td><td>-1</td><td>Outside the nucleus</td><td>0.0005 (-1/1836)</td></tr></tbody></table>			Subatomic particle	Symbol	Relative charge	Location	Mass	Proton	p	+1	Inside the nucleus	1	Neutron	n	0	Inside the nucleus	1	Electron	e <sup>-</sup>	-1	Outside the nucleus	0.0005 (-1/1836)
Subatomic particle	Symbol	Relative charge	Location	Mass																			
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Electron	e <sup>-</sup>	-1	Outside the nucleus	0.0005 (-1/1836)																			

# Lesson Plan Chapter 22 Subatomic

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