

Laboratory Exercise: Double Replacement Reactions...

TABLE II: SCALE UP OF PRECIPITATION REACTIONS

Run #	Chemical Formula of Reagent	Chemical Formula of Reagent	List the four ions present in reagents		insoluble	soluble
					Formula of Solid Product appearance in test tube	Formula of Aqueous Product appearance in test tube
2	NaOH	FeCl <sub>3</sub>	Na <sup>+</sup>	OH <sup>-</sup>	Fe(OH) <sub>3</sub> very small orange	very cloudy orange
			Fe <sup>+3</sup>	Cl <sup>-</sup>		
3	NaNO <sub>3</sub>	FeCl <sub>3</sub>	Na <sup>+</sup>	NO <sub>3</sub> <sup>-</sup>	solid white powder	yellow cloudy
			Fe <sup>+3</sup>	Cl <sup>-</sup>		
5	Na <sub>2</sub> CO <sub>3</sub>	FeCl <sub>3</sub>	Na <sup>+</sup>	CO <sub>3</sub> <sup>-2</sup>	solid orange powdery	very cloudy orange
			Fe <sup>+3</sup>	Cl <sup>-1</sup>		
6	Na <sub>2</sub> PO <sub>4</sub>	FeCl <sub>3</sub>	Na <sup>+</sup>	PO <sub>4</sub> <sup>-3</sup>	yellowish white small	yellow cloudy
			Fe <sup>+3</sup>	Cl <sup>-1</sup>		
12	AgNO <sub>3</sub>	NaOH			very small brown	cloudy brown
13	NaOH	CuSO <sub>4</sub>	N		gel like chunk	blue uncent
17	AgNO <sub>3</sub>	Na <sub>2</sub> CO <sub>3</sub>			white flake small	white cloudy
18	AgNO <sub>3</sub>	Na <sub>2</sub> PO <sub>4</sub>			yellow egg	clear yellow
19	CuSO <sub>4</sub>	Na <sub>2</sub> CO <sub>3</sub>			small blue/white powder	cloudy blue
20	CuSO <sub>4</sub>	Na <sub>3</sub> PO <sub>4</sub>			very small blue particles	cloudy light blue

# Lab 9 2 Double Replacement Reactions Answers

**Yan Bai**



## Lab 9 2 Double Replacement Reactions Answers:

*Chemistry* Eugene LeMay, Jr., Herbert Beall, Karen M. Robblee, Douglas C. Brower, Prentice-Hall Staff, 2002-02

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert

Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day

**Scientific and Technical Aerospace Reports**, 1991 **Regents Chemistry--Physical Setting Power Pack Revised**

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**Prentice Hall Science Explorer** Michael J. Padilla, 2002 **Prentice Hall**

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## **Table of Contents Lab 9 2 Double Replacement Reactions Answers**

1. Understanding the eBook Lab 9 2 Double Replacement Reactions Answers
  - The Rise of Digital Reading Lab 9 2 Double Replacement Reactions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Lab 9 2 Double Replacement Reactions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Lab 9 2 Double Replacement Reactions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Lab 9 2 Double Replacement Reactions Answers
  - Personalized Recommendations
  - Lab 9 2 Double Replacement Reactions Answers User Reviews and Ratings
  - Lab 9 2 Double Replacement Reactions Answers and Bestseller Lists
5. Accessing Lab 9 2 Double Replacement Reactions Answers Free and Paid eBooks
  - Lab 9 2 Double Replacement Reactions Answers Public Domain eBooks
  - Lab 9 2 Double Replacement Reactions Answers eBook Subscription Services
  - Lab 9 2 Double Replacement Reactions Answers Budget-Friendly Options

6. Navigating Lab 9 2 Double Replacement Reactions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Lab 9 2 Double Replacement Reactions Answers Compatibility with Devices
  - Lab 9 2 Double Replacement Reactions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Lab 9 2 Double Replacement Reactions Answers
  - Highlighting and Note-Taking Lab 9 2 Double Replacement Reactions Answers
  - Interactive Elements Lab 9 2 Double Replacement Reactions Answers
8. Staying Engaged with Lab 9 2 Double Replacement Reactions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Lab 9 2 Double Replacement Reactions Answers
9. Balancing eBooks and Physical Books Lab 9 2 Double Replacement Reactions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Lab 9 2 Double Replacement Reactions Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Lab 9 2 Double Replacement Reactions Answers
  - Setting Reading Goals Lab 9 2 Double Replacement Reactions Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lab 9 2 Double Replacement Reactions Answers
  - Fact-Checking eBook Content of Lab 9 2 Double Replacement Reactions Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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