

Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	Ion Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	F^-	9	10	-1	Anion
Iodine	I^-	53	54	-1	Anion
Sulfur	S^{2-}	16	18	-2	Anion
Potassium	K^+	19	18	+1	Cation
Calcium	Ca^{2+}	20	18	+2	Cation
Bromine	Br^-	35	36	-1	Anion
Strontium	Sr^{2+}	38	36	+2	Cation
Oxygen	O^{2-}	8	10	-2	Anion
Magnesium	Mg^{2+}	12	10	+2	Cation
Aluminum	Al^{3+}	13	10	+3	Cation
Selenium	Se^{2-}	34	36	-2	Anion
Lithium	Li^+	3	2	+1	Cation
Rubidium	Rb^+	37	36	+1	Cation
Chlorine	Cl^-	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & nonmetal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.



5. What is the formula for the compound formed in question #4?



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Table of Contents Ions Packet Answers

1. Understanding the eBook Ions Packet Answers
 - The Rise of Digital Reading Ions Packet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ions Packet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ions Packet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ions Packet Answers
 - Personalized Recommendations
 - Ions Packet Answers User Reviews and Ratings
 - Ions Packet Answers and Bestseller Lists

5. Accessing Ions Packet Answers Free and Paid eBooks
 - Ions Packet Answers Public Domain eBooks
 - Ions Packet Answers eBook Subscription Services
 - Ions Packet Answers Budget-Friendly Options
6. Navigating Ions Packet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ions Packet Answers Compatibility with Devices
 - Ions Packet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ions Packet Answers
 - Highlighting and Note-Taking Ions Packet Answers
 - Interactive Elements Ions Packet Answers
8. Staying Engaged with Ions Packet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ions Packet Answers
9. Balancing eBooks and Physical Books Ions Packet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ions Packet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ions Packet Answers
 - Setting Reading Goals Ions Packet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ions Packet Answers
 - Fact-Checking eBook Content of Ions Packet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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