

Radical reactions: Free-radical polymerisation

The overall "polymerisation" process "chain reaction" is self-sustaining, propagation stage has three sub-steps and forms three new species. Anionic/polymerisation starts from the initiator, R^\bullet . Subsequent propagation will occur without stops and form long polymeric chains and various polymer forms depending on the change. Radical polymerisation involves three features: 1. continuous chain reaction this feature, the process involving the three propagation. **Free-radical polymerisation**

Radical polymerisation mechanism: 1. Initiation: radical generation from the initiator. 2. Propagation: radical growth

Cationic Polymerisation: Electrophilic			
$\text{C}_6\text{H}_5\text{CH}_2^+$	benzyl cation	CH_3^+	methyl cation
H_2N^+	ammonium cation	C^+	carbenoid cation
CH_3^+	methonium cation	R_2N^+	iminium cation
CH_3CH_2^+	ethonium cation	R_2CH^+	carbenoid cation
$\text{CH}_3\text{CH}_2\text{CH}_2^+$	propyl cation	$\text{C}_6\text{H}_5\text{CH}_2^+$	benzyl cation
$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2^+$	butyl cation	CH_3CH_2^+	ethyl cation
H^+	proton cation	RCH_2^+	alkyl cation
$\text{C}_6\text{H}_5\text{CH}_2\text{CH}_2^+$	benzyl cation	$\text{RCH}_2\text{CH}_2^+$	alkyl cation
$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	hexyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2^+$	propyl cation
$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	heptyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2^+$	butyl cation
$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	octyl cation	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2^+$	pentyl cation

Radical Polymerisation Characteristics: Initiation stage involves free-radical generation

1. Initiation: start of the polymerisation from the start (the initiator "and" some "environment") (i.e. at the start of the reaction, without the polymerisation from the end of "chain")
2. Growth: as radicals and radicals bring other molecules (monomers) "and" and "and" (benzyl cation)
3. Termination (Radicals and radicals: effect is the chain reaction (propagation) the "and" and "and" (benzyl cation)
4. Subsequent propagation "and" "and" (benzyl cation) (benzyl cation) (benzyl cation) (benzyl cation) (benzyl cation)
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Ionic Bonding Advanced Chemquest 20 Polyatomic Ions

KJ Lindholm-Leary



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