

Intro to Acids and Bases

Name: _____
 Class: _____
 Date: _____

Define each term: Definitions of acids and bases:

Arrhenius Definitions:

- acid substances that release dissolved hydrogen ions (H^+) (acids H^+ gases formed in gases as the hydrogens form H_2 and acid H^+ form H_2O) and conjugated anions (A^-)
- base substances which release hydroxide ions (OH^-)

Brønsted-Lowry Definitions:

- acid substances that donate a proton (H^+) (acid donate)
- base substances which accept a proton (H^+) (base receive)

Table 1: Representative reactions for H^+ (acid) and OH^- (base) with some representative substances.

Acids	Bases
1. $HCl(aq) + H_2O(l) \rightarrow H_3O^+(aq) + Cl^-(aq)$	1. $NaOH(aq) + H^+(aq) \rightarrow H_2O(l) + Na^+(aq)$
2. $HNO_3(aq) + H_2O(l) \rightarrow H_3O^+(aq) + NO_3^-(aq)$	2. $Ca(OH)_2(aq) + 2H^+(aq) \rightarrow 2H_2O(l) + Ca^{2+}(aq)$
3. $H_2SO_4(aq) + H_2O(l) \rightarrow H_3O^+(aq) + HSO_4^-(aq)$	3. $Al(OH)_3(s) + 3H^+(aq) \rightarrow Al^{3+}(aq) + 3H_2O(l)$
4. $H_2CO_3(aq) + H_2O(l) \rightarrow H_3O^+(aq) + HCO_3^-(aq)$	4. $Fe(OH)_3(s) + 3H^+(aq) \rightarrow Fe^{3+}(aq) + 3H_2O(l)$
5. $H_2O(l) + H^+(aq) \rightarrow H_3O^+(aq)$	5. $OH^-(aq) + H^+(aq) \rightarrow H_2O(l)$

Critical Thinking Questions

1. List all the substances in Table 1 that are Arrhenius acids.

2. List all the substances in Table 1 that are Arrhenius bases.

3. List all the substances in Table 1 that are Brønsted-Lowry acids.

4. List all the substances in Table 1 that are Brønsted-Lowry bases.

5. List all the substances in Table 1 that are both acids and bases.

6. List all the substances in Table 1 that are both acids and bases.

7. List all the substances in Table 1 that are both acids and bases.

8. List all the substances in Table 1 that are both acids and bases.

9. List all the substances in Table 1 that are both acids and bases.

10. List all the substances in Table 1 that are both acids and bases.

11. List all the substances in Table 1 that are both acids and bases.

12. List all the substances in Table 1 that are both acids and bases.

13. List all the substances in Table 1 that are both acids and bases.

14. List all the substances in Table 1 that are both acids and bases.

15. List all the substances in Table 1 that are both acids and bases.

16. List all the substances in Table 1 that are both acids and bases.

17. List all the substances in Table 1 that are both acids and bases.

18. List all the substances in Table 1 that are both acids and bases.

19. List all the substances in Table 1 that are both acids and bases.

20. List all the substances in Table 1 that are both acids and bases.

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Arturo Cuomo



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Table of Contents Intro To Acids And Bases Chemquest 48

1. Understanding the eBook Intro To Acids And Bases Chemquest 48
 - The Rise of Digital Reading Intro To Acids And Bases Chemquest 48
 - Advantages of eBooks Over Traditional Books
2. Identifying Intro To Acids And Bases Chemquest 48
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Intro To Acids And Bases Chemquest 48
 - User-Friendly Interface
4. Exploring eBook Recommendations from Intro To Acids And Bases Chemquest 48
 - Personalized Recommendations
 - Intro To Acids And Bases Chemquest 48 User Reviews and Ratings
 - Intro To Acids And Bases Chemquest 48 and Bestseller Lists
5. Accessing Intro To Acids And Bases Chemquest 48 Free and Paid eBooks
 - Intro To Acids And Bases Chemquest 48 Public Domain eBooks
 - Intro To Acids And Bases Chemquest 48 eBook Subscription Services
 - Intro To Acids And Bases Chemquest 48 Budget-Friendly Options
6. Navigating Intro To Acids And Bases Chemquest 48 eBook Formats
 - ePub, PDF, MOBI, and More
 - Intro To Acids And Bases Chemquest 48 Compatibility with Devices
 - Intro To Acids And Bases Chemquest 48 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Intro To Acids And Bases Chemquest 48
 - Highlighting and Note-Taking Intro To Acids And Bases Chemquest 48
 - Interactive Elements Intro To Acids And Bases Chemquest 48
8. Staying Engaged with Intro To Acids And Bases Chemquest 48

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Intro To Acids And Bases Chemquest 48
- 9. Balancing eBooks and Physical Books Intro To Acids And Bases Chemquest 48
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Intro To Acids And Bases Chemquest 48
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Intro To Acids And Bases Chemquest 48
 - Setting Reading Goals Intro To Acids And Bases Chemquest 48
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Intro To Acids And Bases Chemquest 48
 - Fact-Checking eBook Content of Intro To Acids And Bases Chemquest 48
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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