CAPS-17

#### SCQ (Single Correct Type):

Nuclei of radioactive element A are produced at rate 't' at any time t. The element A has 1. decay constant ). Let N be the number of nuclei of element A at any time t. At time  $t = t_0$ ,  $\frac{dN}{dt}$ is minimum. Then the number of nuclei of element A at time  $t = t_0$  is

$$(A) \frac{\lambda t_0^2 - 2t_0}{\lambda^2}$$

(B) 
$$\frac{\mathbf{t}_0 - \lambda \mathbf{t}_0^2}{\lambda^2}$$

(A) 
$$\frac{\lambda t_0^2 - 2t_0}{\lambda^2}$$
 (B)  $\frac{t_0 - \lambda t_0^2}{\lambda^2}$  (C)  $\frac{2t_0 - \lambda t_0^2}{\lambda}$ 

- (D) None of these
- 2. A radioactive sample decays by β-emission. In first two seconds 'n' β-particles are emitted and in next 2 seconds, '0.25n' β-particles are emitted. The half life of radioactive nuclei is
  - (A) 2 sec.

(B) 4 sec.

(C) 1 sec

- (D) none
- A factory produces a radionuclide at a constant rate R. Initially there were no nuclide present. 3. Then activity of the sample produced by factory just after an average life of nuclide passes away is:
  - (A) R

(B) Re\*

- (C)  $R(1-e^{-t})$  (D)  $\frac{R}{2}(1-e^{-t})$
- 4. A radionuclide with decay constant 3, is produced in a reactor at a constant rate \alpha nuclei per second. During each decay energy E<sub>o</sub> MeV is released. If the production of radio nuclides started at t = 0, total energy released upto time t is given by equation

- (A)  $E_{0} \propto (1 e^{-\lambda t})$  (B)  $E_{0} \frac{\alpha}{\lambda} (1 e^{-\lambda t})$  (C)  $E_{0} (\alpha t \frac{\alpha}{\lambda} e^{-\lambda t})$  (D)  $E_{0} \left[ \alpha t \frac{\alpha}{\lambda} (1 e^{-\lambda t}) \right]$
- The count rate observed from a radioactive source at 't' second was No and at 4t second it 5. was  $\frac{N_{\rm o}}{16}$  . The count rate observed, at  $\left(\frac{11}{2}\right)t$  second will be
  - (A) No

- (B)  $\frac{N_o}{l}$  (C)  $\frac{N_o}{l}$

- (D) None of these
- 6. The count rate from 100 cm3 of a radio-active liquid is C. Some of this liquid is now discarded. The count rate of the remaining liquid is found to be C/10 after three half-lives. The volume of the remaining liquid in cm3 is
  - (A) 20

(B) 40

(C) 60

(D) 80

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- 1. Understanding the eBook Grade 1exam Papers Physics Caps Exemplar
  - The Rise of Digital Reading Grade 1exam Papers Physics Caps Exemplar
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade 1exam Papers Physics Caps Exemplar
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade 1exam Papers Physics Caps Exemplar
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade 1exam Papers Physics Caps Exemplar
  - Personalized Recommendations
  - Grade 1exam Papers Physics Caps Exemplar User Reviews and Ratings

- Grade 1exam Papers Physics Caps Exemplar and Bestseller Lists
- 5. Accessing Grade 1exam Papers Physics Caps Exemplar Free and Paid eBooks
  - o Grade 1exam Papers Physics Caps Exemplar Public Domain eBooks
  - Grade 1exam Papers Physics Caps Exemplar eBook Subscription Services
  - Grade 1exam Papers Physics Caps Exemplar Budget-Friendly Options
- 6. Navigating Grade 1exam Papers Physics Caps Exemplar eBook Formats
  - o ePub, PDF, MOBI, and More
  - Grade 1exam Papers Physics Caps Exemplar Compatibility with Devices
  - Grade 1exam Papers Physics Caps Exemplar Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Grade 1exam Papers Physics Caps Exemplar
  - Highlighting and Note-Taking Grade 1exam Papers Physics Caps Exemplar
  - Interactive Elements Grade 1exam Papers Physics Caps Exemplar
- 8. Staying Engaged with Grade 1exam Papers Physics Caps Exemplar
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Grade 1exam Papers Physics Caps Exemplar
- 9. Balancing eBooks and Physical Books Grade 1exam Papers Physics Caps Exemplar
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Grade 1exam Papers Physics Caps Exemplar
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 1exam Papers Physics Caps Exemplar
  - Setting Reading Goals Grade 1exam Papers Physics Caps Exemplar
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 1exam Papers Physics Caps Exemplar
  - Fact-Checking eBook Content of Grade 1exam Papers Physics Caps Exemplar
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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